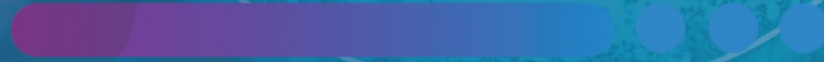




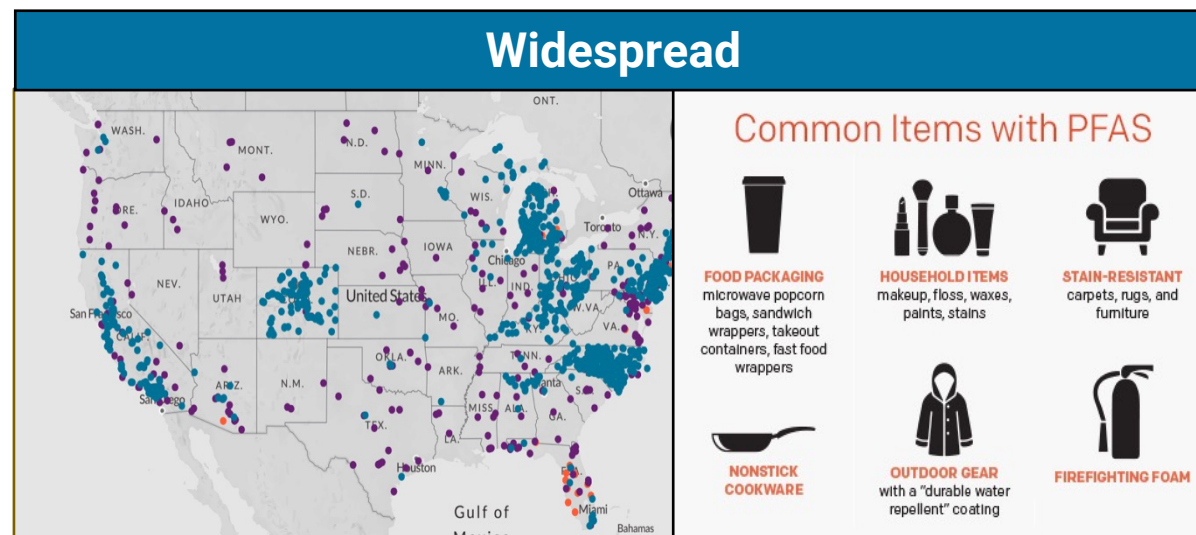
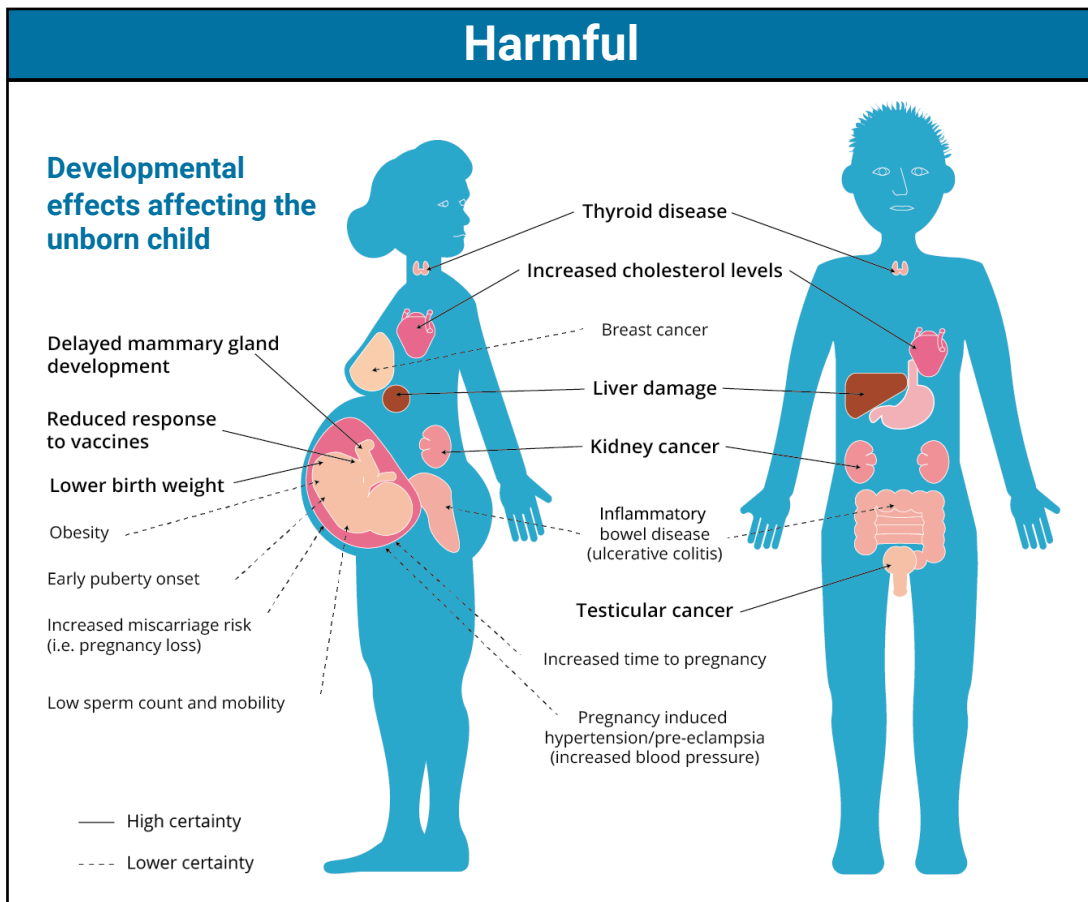
Electrochemical Destruction of of PFAS in High Strength Wastes



Sophie Waterhouse

January 22, 2024

NEWEA 2024 Annual Conference and Exhibit

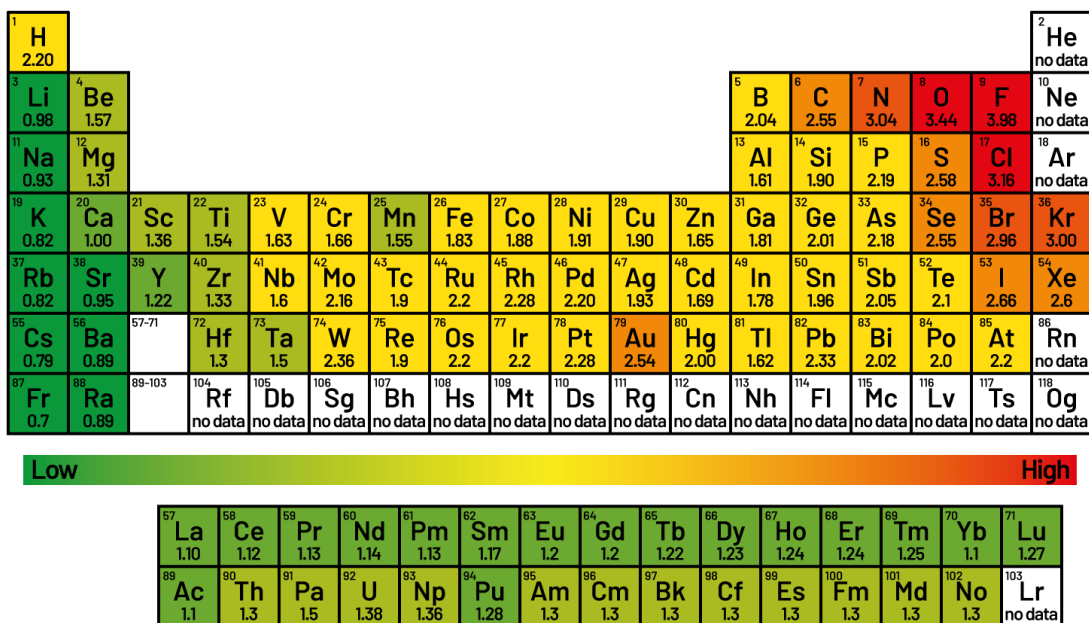


How it Works

Electrochemical Oxidation 101



Electronegativity of the Elements



(Helmenstine, 2021)

PFAS	E ⁰ anion (V/SHE)	E ⁰ acid (V/SHE)
PFBA	2.96	3.65
PFHxA	2.97	3.70
PFHpA	2.96	3.58
PFOA	2.91	3.71
PFNA	3.12	3.76
PFBS	3.71	3.96
PFHxS	3.72	4.00
PFOS	3.74	4.02

Source: Radjenovic et al., ES&T, 2020

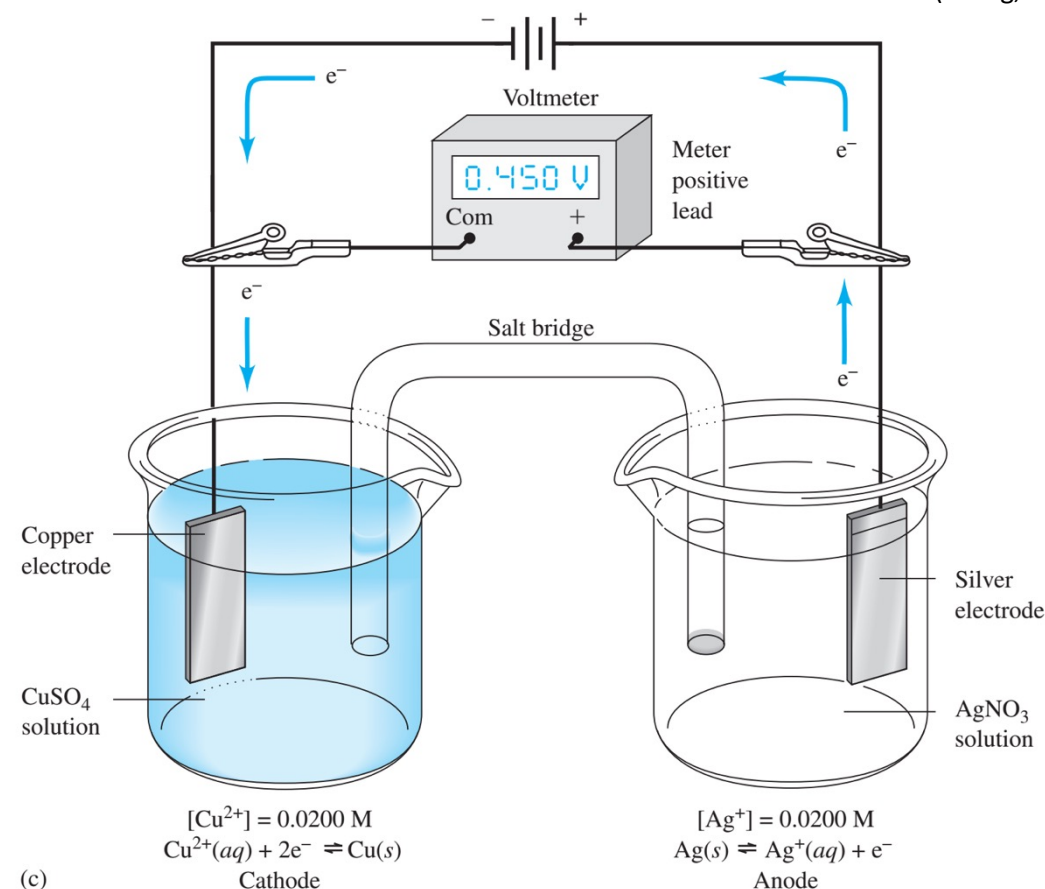
E⁰ of OH• = 2.59(V/SHE)

(Skoog, 2005)

Destructive Technologies for PFAS:

- Electrochemical Oxidation (EOx)
- Supercritical Water Oxidation
- Plasma Oxidation
- Hydrothermal Alkaline Treatment
- Reductive Defluorination
- Photochemical

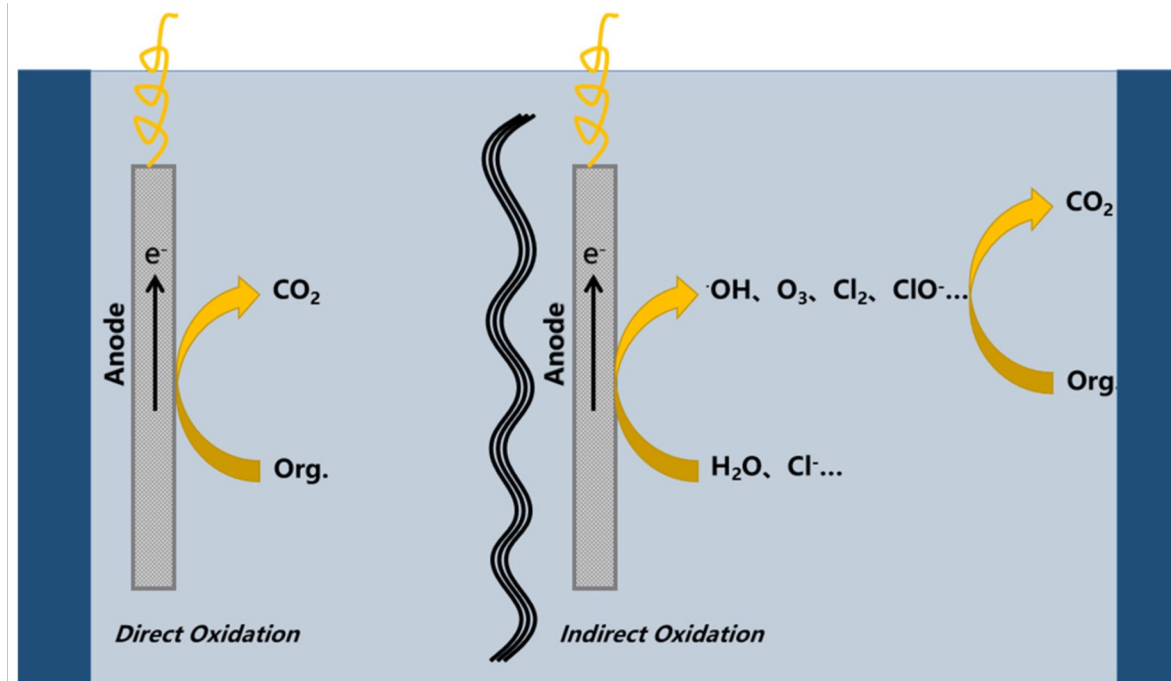
Overpotential at anode EOx allows for destruction of PFAS.



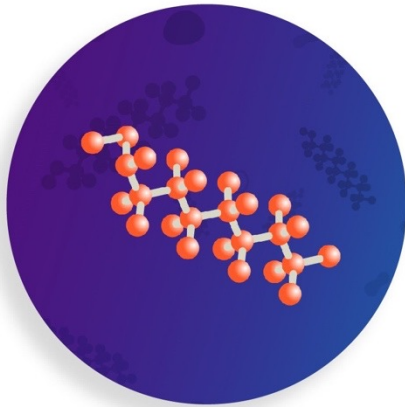
(1) Direct Oxidation: Potential (V) at anode greater than E^0 of PFAS. Anode “takes” e^- from PFAS resulting in $PFAS^\bullet$ and other oxidized forms.

(2) Indirect Oxidation: Anode “takes” electron from other constituents in the water (e.g., OH^-), generating radicals *in situ* (e.g. OH^\bullet). Oxidizes $PFAS^\bullet$ and other pollutants at relatively low power.

Both mechanisms combine for PFAS destruction!

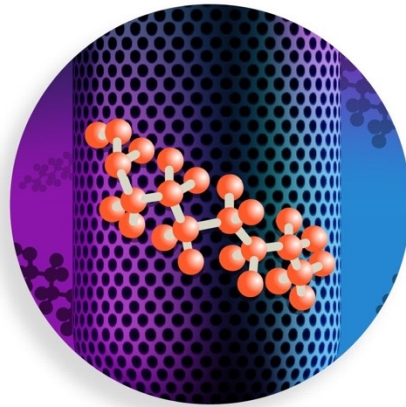


Direct and Indirect Oxidation, (Liu et al., 2022)



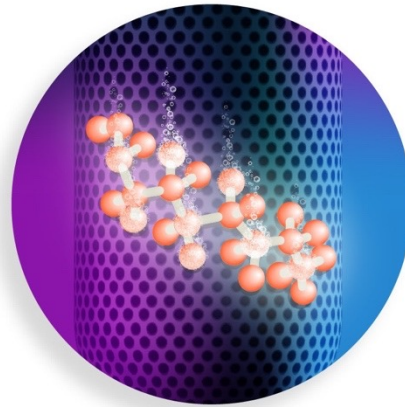
1

PFAS in liquid waste



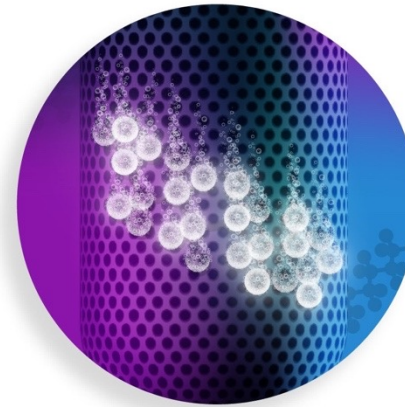
2

The PFAS are adsorbed onto anode surface



3

Free electrons break C-F bonds resulting in CO₂, HF, F⁻



4

Oxidant radicals generated through electrochemical process



5

PFAS are permanently destroyed

Present Focus

- Landfill leachates
- AFFF
- Foam Fractionate
- RO/NF concentrate

Under Investigation

- Direct treatment of IX Resin

50% of PFAS in the environment is sourced from landfills

PFOA and PFOS as hazardous compounds via CERCLA “soon”



Mobile Bench System
Batch Treatment



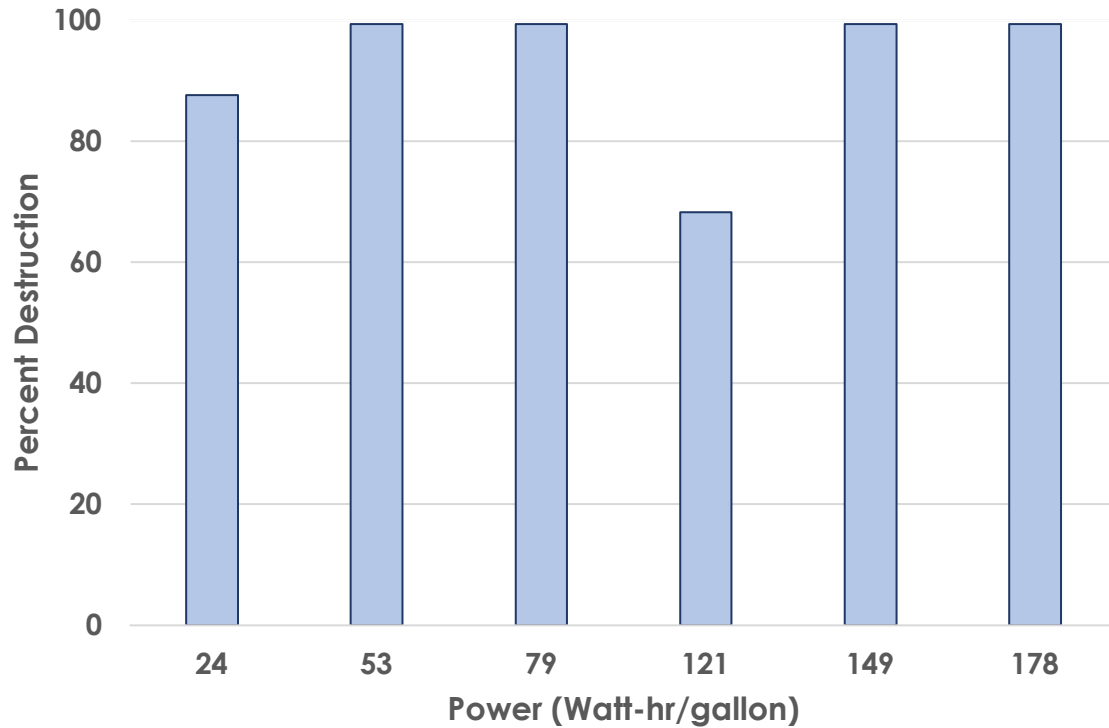
Full-Scale Reactor Field
Deployment System
Continuous Flow



Multi-Reactor, Multi-Skid Installation
Destruction as a Service™
Permanent Install

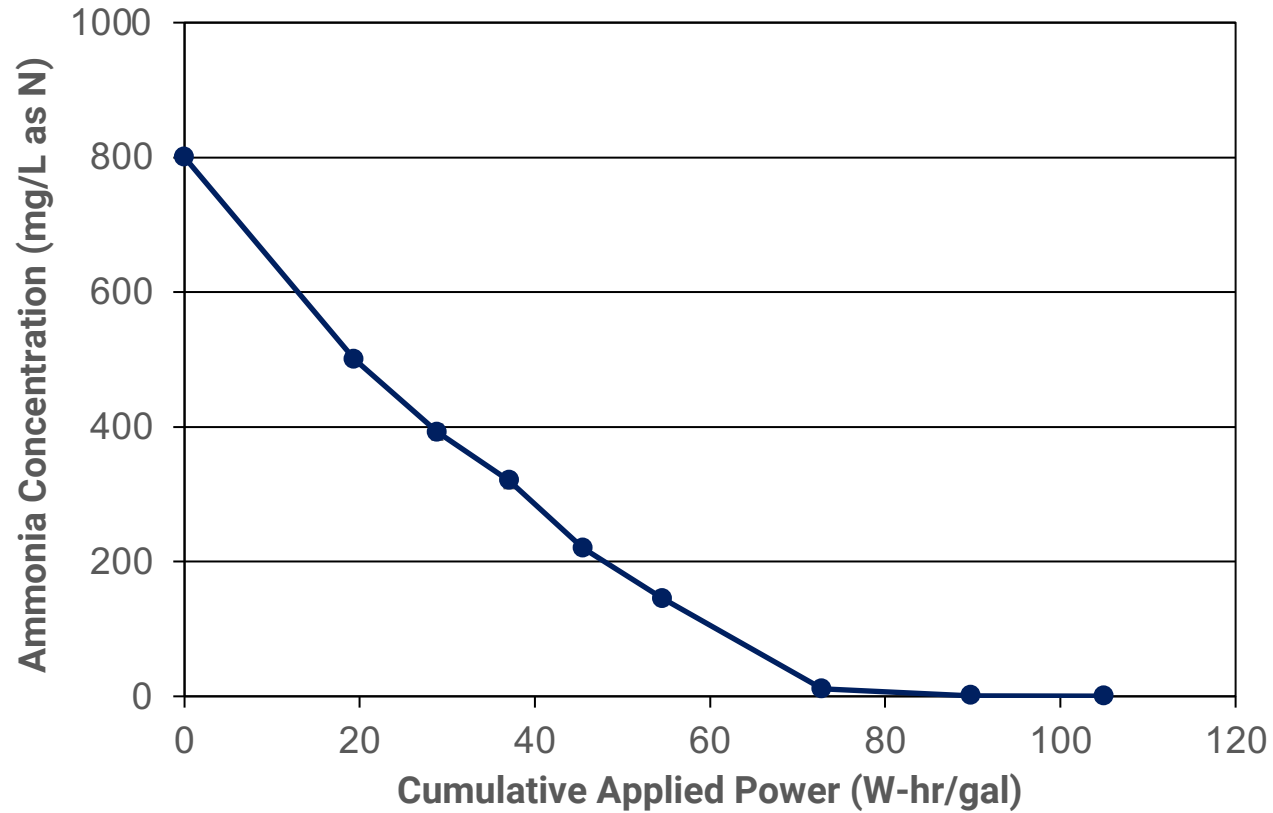
Bench Testing Results

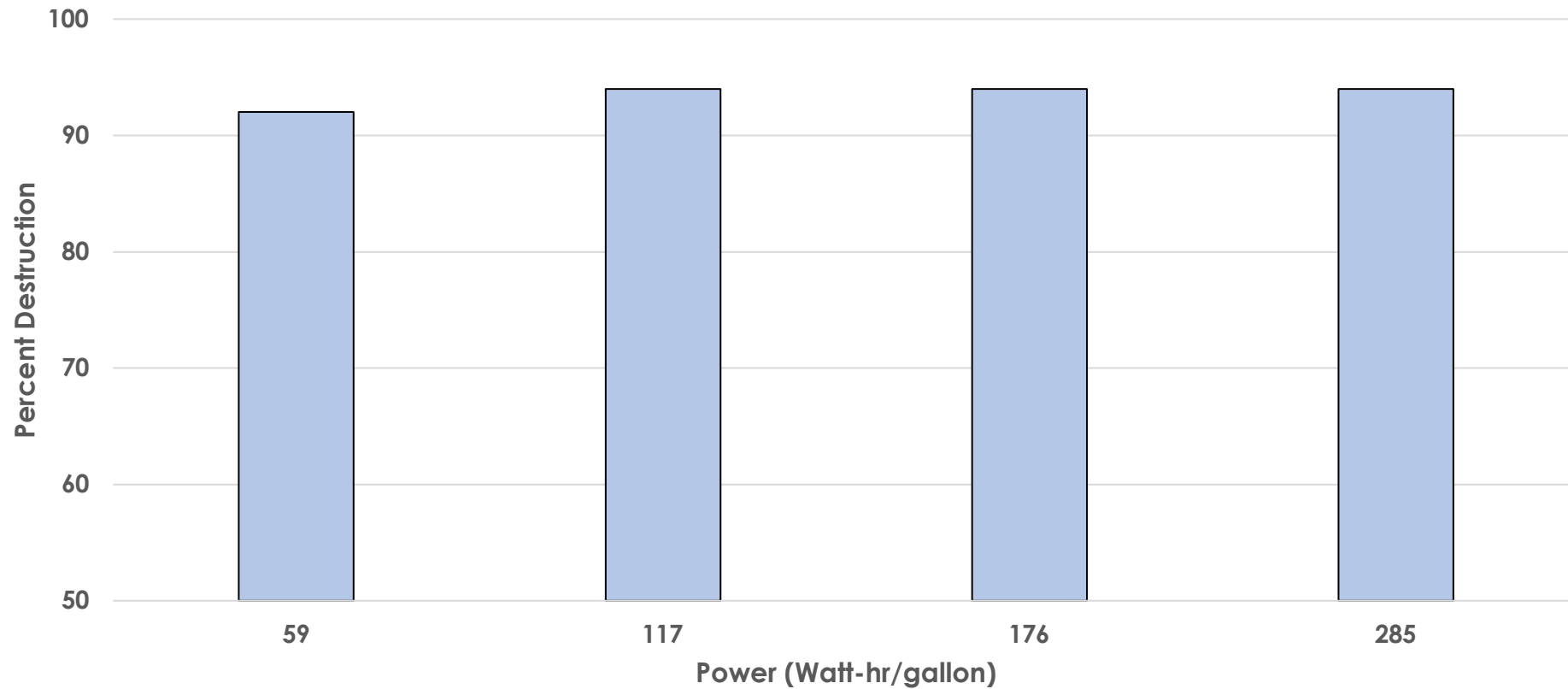
Leachate, foam fractionate, & more



Compound	% Destruction at 53 watt-hr/gallon	Decay Rate (min ⁻¹)
PFOA	99.9%	0.024
PFOS	96.6%	0.012
PFNA	95.2%	0.011
PFHxS	98.4%	0.015
PFDA	95.9%	0.012
PFHpA	99.4%	0.019

Starting PFAS6 Concentration: 11,000 ppt





Starting PFAS6 Concentration: 520,000 ppt



Recent Field Testing Results

In the field with production ready reactors

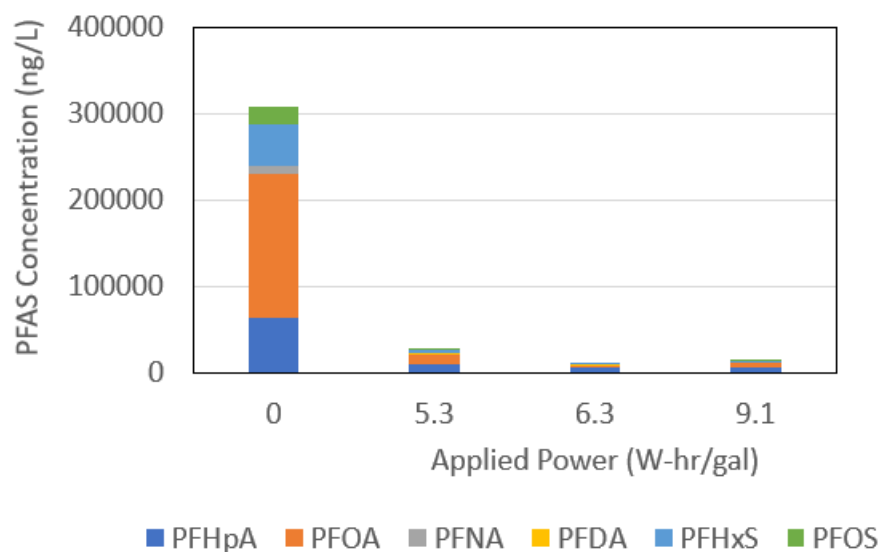
- Type: Landfill
 - Tested five leachates and RO Concentrate

Parameter (units)	Value
Aluminum (mg/L)	0.773
Calcium (mg/L)	847
Chromium (mg/L)	0.637
Iron (mg/L)	54.8
Magnesium (mg/L)	829
Manganese (mg/L)	4.9
Ammonia (mg/L as N)	20.7
COD (mg/L)	8620
NO ₂ +NO ₃ (mg/L as N)	626
Total Nitrogen (mg/L as N)	1110
TDS (mg/L)	50900
TKN (mg/L as N)	482
TOC (mg/L)	2960
UV ₂₅₄ (1/cm)	88

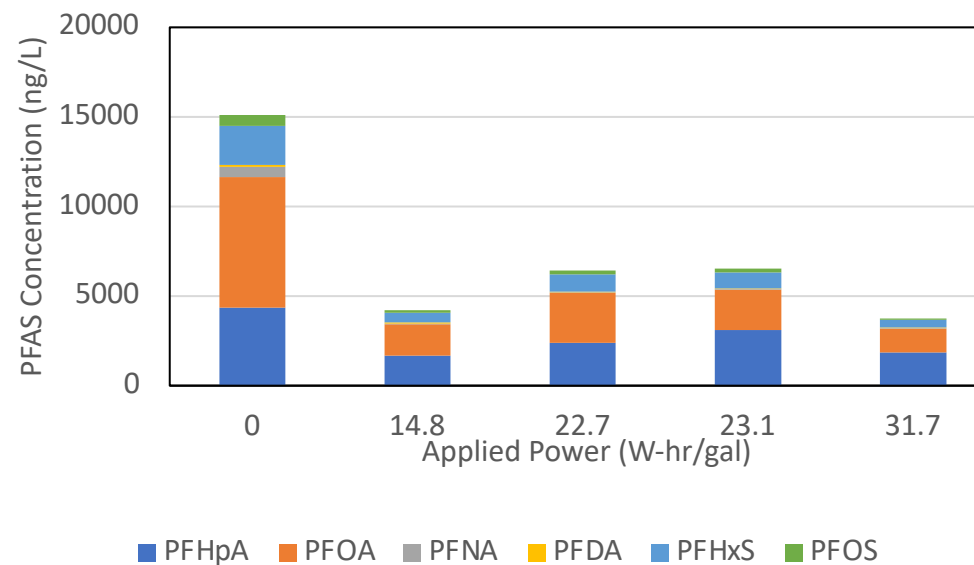


Destroys PFAS with Low Power

Thermal Concentrate from Landfill Leachate – continuous flow



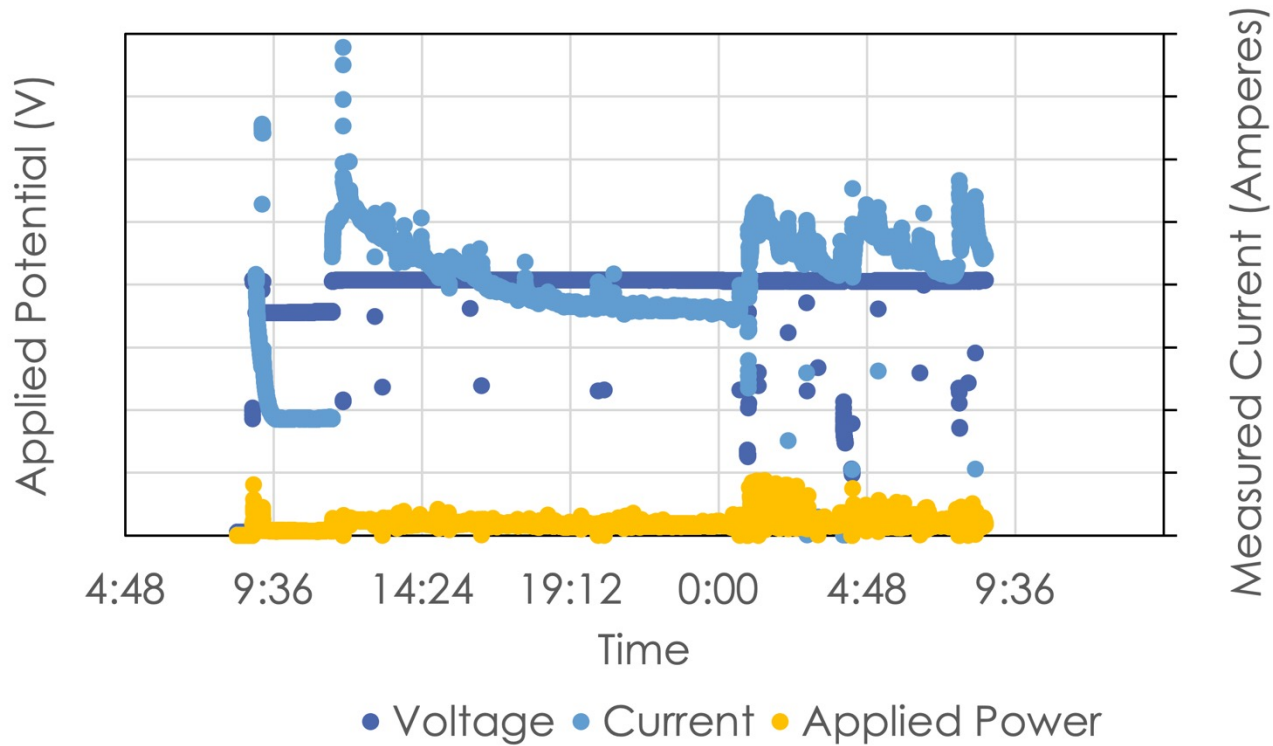
RO Concentrate from Landfill Leachate – continuous flow



DESTRUCTION IN DIFFERENT WATER MATRICES

	Synthetic Brine		Raw Leachate		Thermal Concentrate		Foam Fractionate	
	Initial Conc (ng/L)	Decay Rate (min ⁻¹)	Initial Conc (ng/L)	Decay Rate (min ⁻¹)	Initial Conc (ng/L)	Decay Rate (min ⁻¹)	Initial Conc (ng/L)	Decay Rate (min ⁻¹)
PFOA	21,500	0.020	1,118	0.025	168,000	0.024	4,510,000	0.011
PFOS	33,500	0.024	486	0.022	22,200	>0.02	147,000	0.016
PFDA	---	---	---	---	6,790	>0.0125	43,000	0.017
PFNA	---	---	---	---	8,420	>0.018	96,400	0.014
PFHxS	---	---	---	---	47,000	0.02	50,700	0.008
PFHpA	---	---	---	---	62,700	0.008	9,370	0.0014

24-hour fouling test run. Cleaned with ~2% acid soln.



Fouled Electrodes

Clean Electrodes

- Destruction of long and short chain PFAS compounds
- Continuous flow steady state operation
- Low power (<50 W-hr/gal)
- Ambient temperatures and pressures
- Long lasting electrodes (years)
- Cleaning allows for fouling removal and the recovery of original performance
- Remote operation
- No waste produced



- Future Development
 - Ion Exchange Brine
 - Spent IX Resin
 - Regenerable GAC waste
 - Biosolids



ACLARITY TEAM

Chris Hull

Elisabeth Christ

Orren Schneider

Anil Krosuri

Leanne Hersey

OPERATORS of FACILITIES



THANK YOU!

sophie.waterhouse@aclaritywater.com

Dr. Orren Schneider, Chief Science Officer

orren.schneider@aclaritywater.com

www.aclaritywater.com