

# Decoding the fate and transport of PFAS compounds in sludge undergoing thermal oxidation

May 24, 2022 9:00 AM - 9:30 AM | Presented by Sudhakar Viswanathan, Veolia, Cary, NC

Regional Partners



Conference Host



# Research Team



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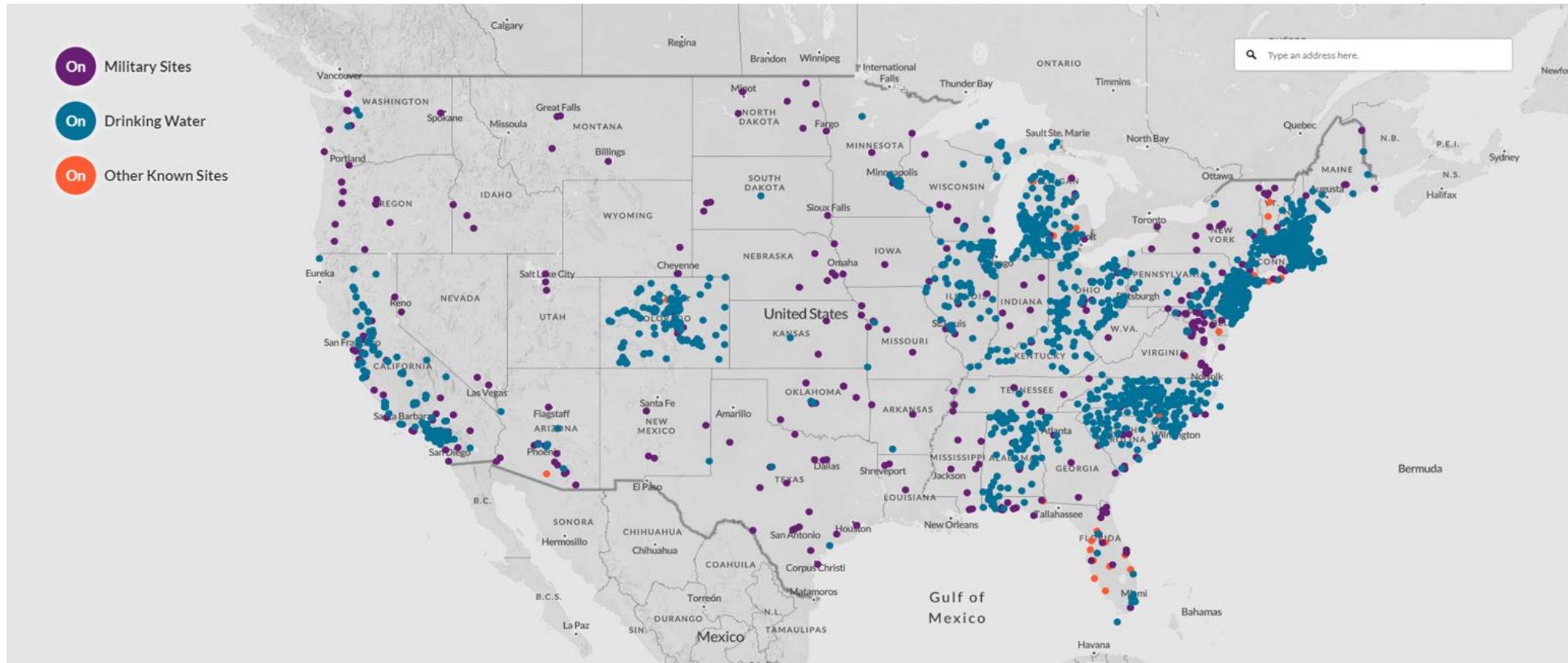
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# What is PFAS?



- Family of forever chemicals
  - ~ 4730 compounds
  - PFOA, PFOS, GenX, TEFLON...
  - Aqueous Film Forming Foam (AFFF)
- Persistent
- Toxic
  - ECD
  - Carcinogenic
- Bioaccumulative
  - It's in all our blood 😞

# Is PFAS in my sludge?



Source: ewg.org

# Local Concerns

## ENVIRONMENT

### Malne waste handlers eye cost concerns from sludge ban that aims to prevent PFAS contamination

BY ANNIE ROPEIK | MAINE  
PUBLISHED 7:00 AM ET MAY. 02, 2022

Maine towns and wastewater plants say they're reeling from the passage of a statewide ban on sludge fertilizer that aims to reduce harmful PFAS chemical contamination in soil and water.

Members of a range of groups that work with these so-called biosolids, from waste management companies to sewage plants and municipalities, met for a forum on the issue hosted by Maine's E2Tech trade group last week.

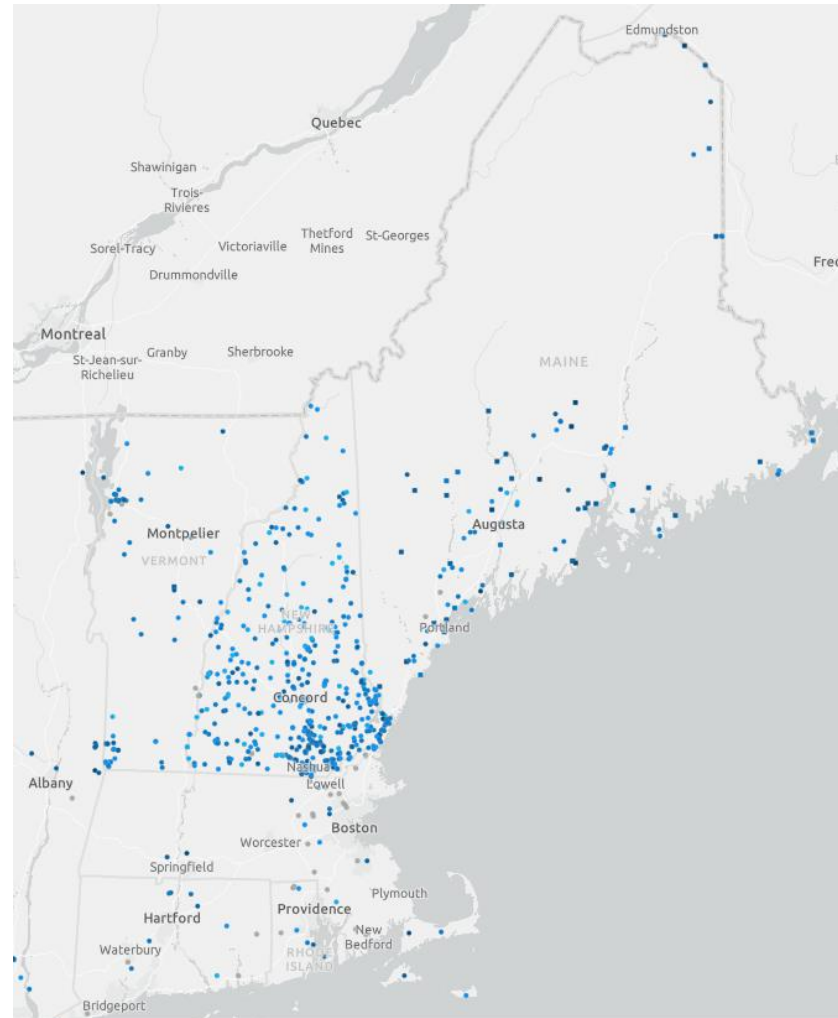
These solids are left over from wastewater and have historically been treated for use as a nutrient-rich fertilizer instead of being landfilled or incinerated. But they can contain high levels of PFAS due to the chemicals' ubiquity in human bodies and other wastewater sources.

## EQUILIBRIUM & SUSTAINABILITY

### New England governors approve transformative laws on 'forever chemicals'

### State task force recommends Mass. do more to crack down on PFAS

By Dharna Noor | Globe Staff. Updated April 20, 2022, 7:10 p.m.



## Legend

### Known Contamination

#### Sites with Water Testing

#### PFAS Level (ng/L)

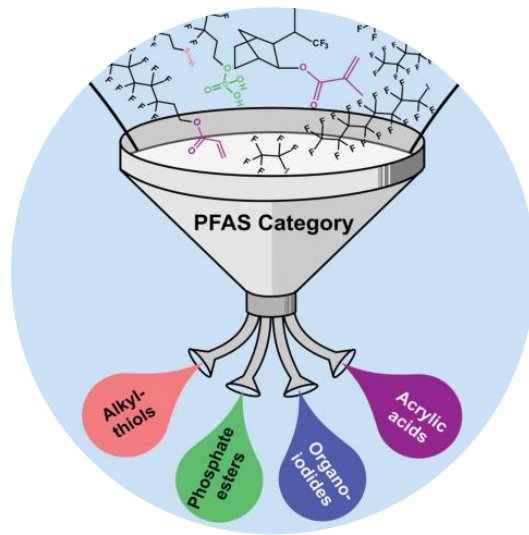
- > 100,000
- > 1,000 - 100,000
- > 70 - 1,000
- > 10 - 70
- 0 - 10
- Unknown

#### Sites with Solids Testing

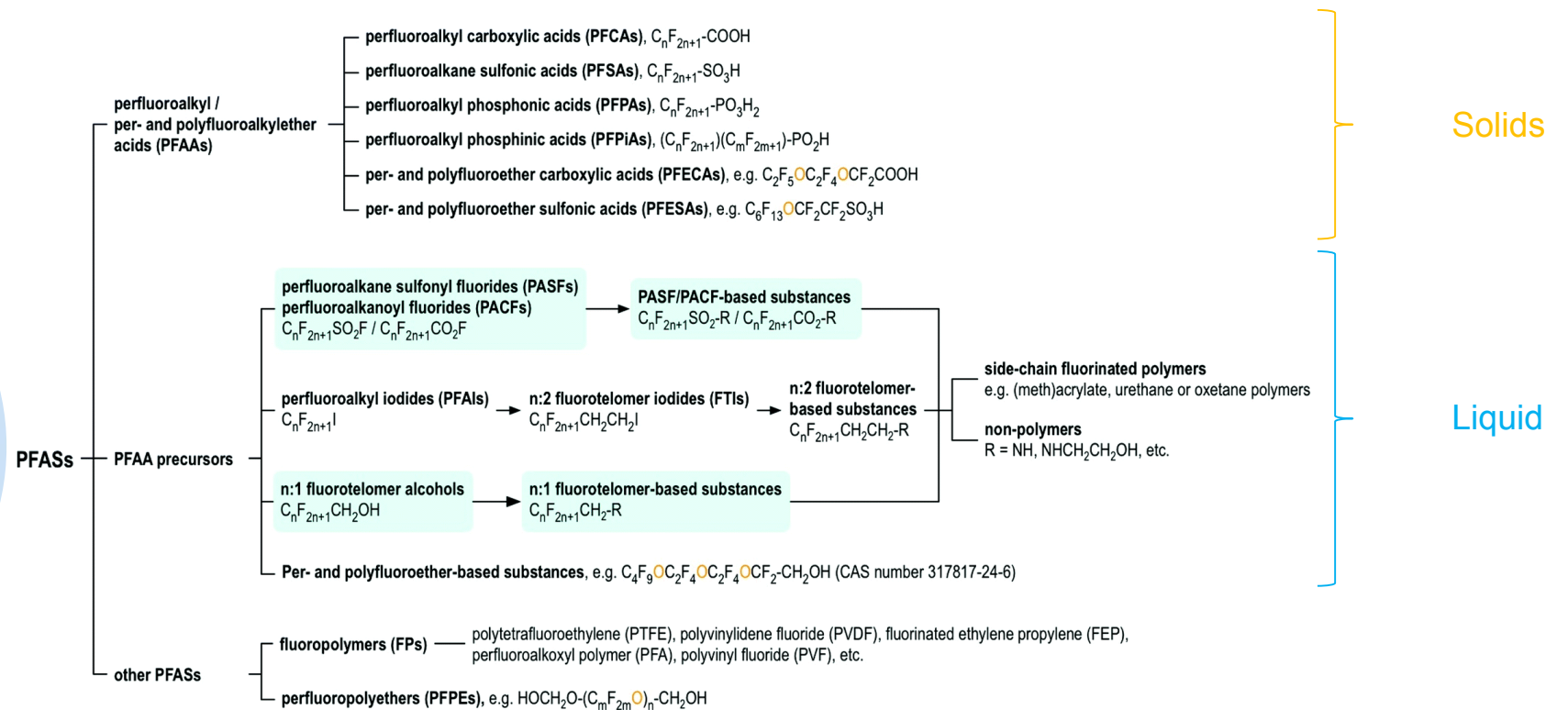
#### PFAS Level (ng/L)

- > 100,000
- > 10,000 - 100,000
- > 100 - 10,000
- 0 - 100
- Unknown

# PFAS in wastewater treatment facility



Source: rsc.li/espi



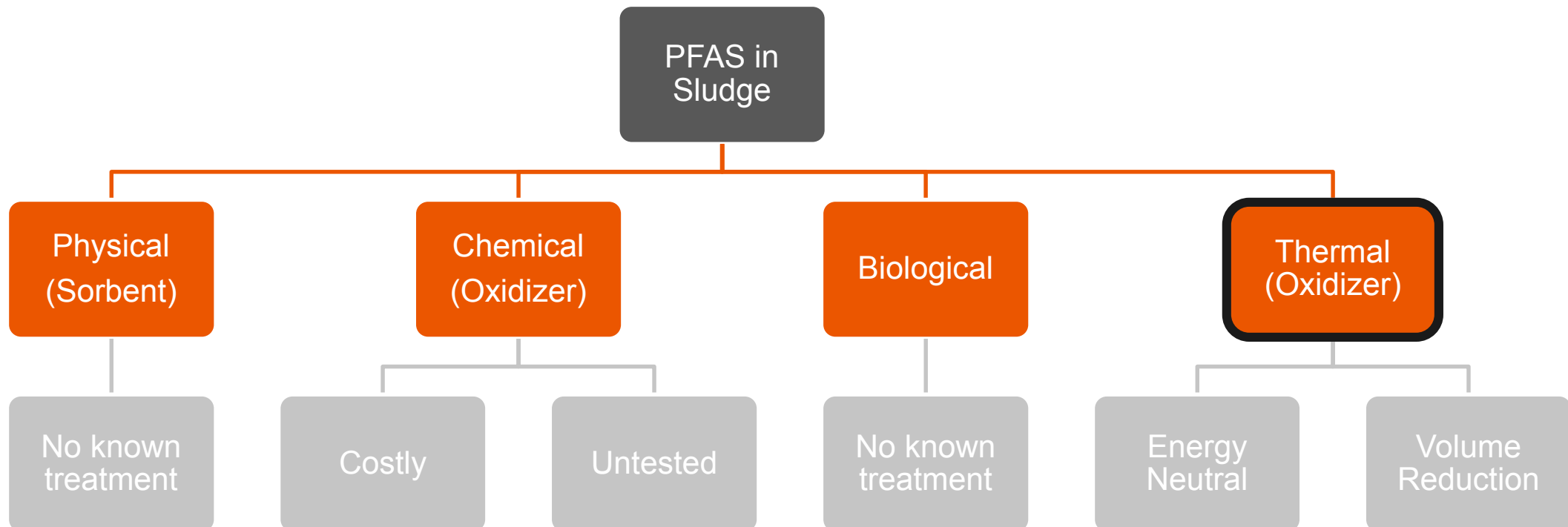
## ● Laboratory Capabilities

- ~40 compounds (2021)
- Total oxidizable precursor (TOP) assay

## ● Further Reading

- Dr. Mouser, Univ. of NH

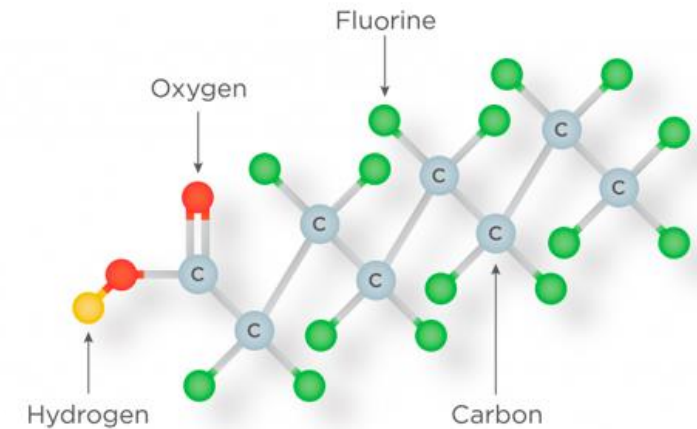
# How to treat PFAS laden sludge?





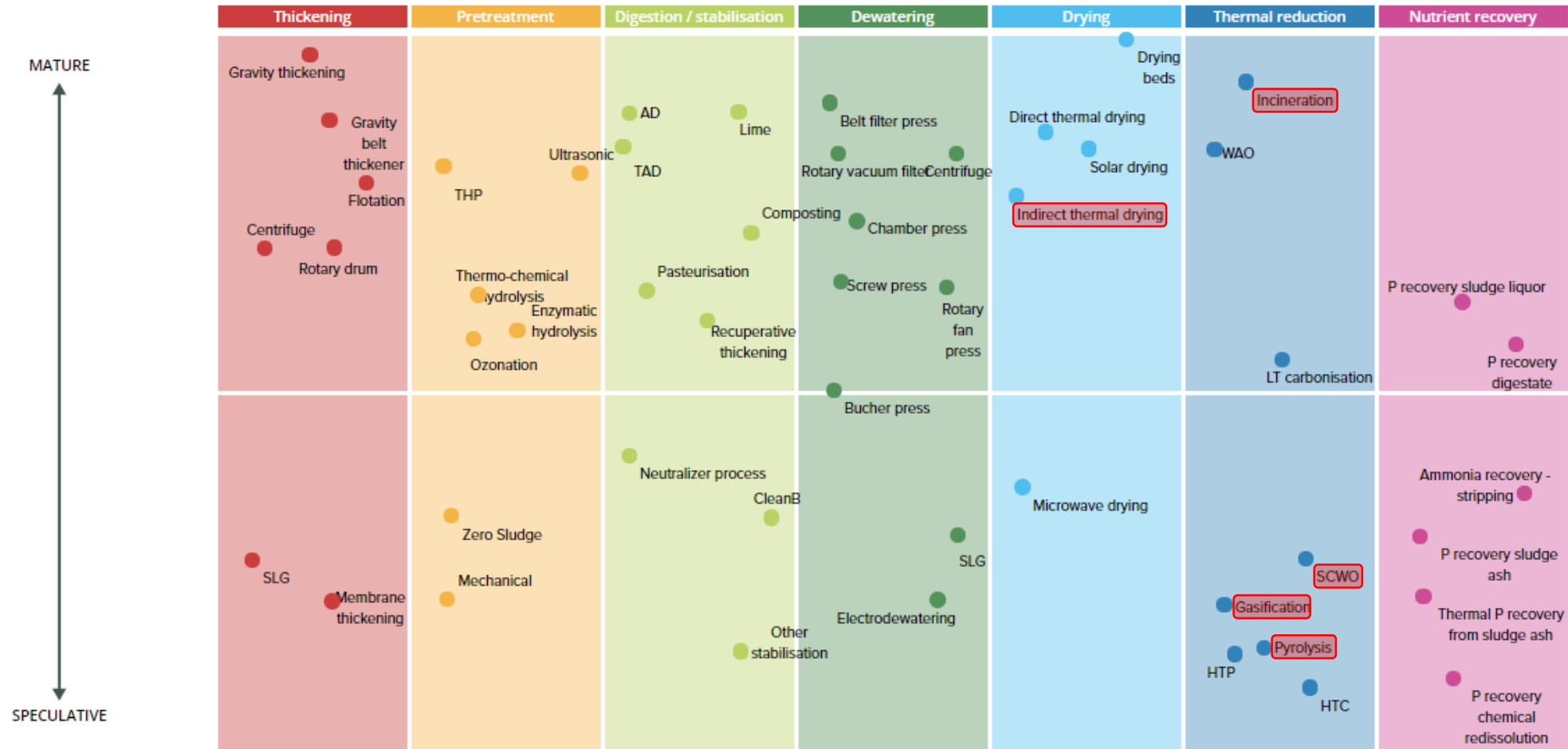
# Thermal treatment, in literature!

- Literature review offered limited information on sludge
  - The knowledge on how PFAS behaves in combustion processes is scarce, but consensus in the limited scientific literature is that degradation of PFOS occurs at temperatures above 500°C. However fluorinated by-products are formed, which in themselves may have undesired properties. A study conducted by the United States Environmental Protection Agency and 3M states that degradation of PFOS occurs at temperatures above 600°C, and the main degradation products are the potent greenhouse gases  $\text{CF}_4$  and  $\text{C}_2\text{F}_6$  (Taylor et al 2003).
  - Thermally treated PFOS-contaminated sludge with an addition of  $\text{Ca}(\text{OH})_2$  (Wang et al 2011; Wang et al. 2013) has reduced the emissions of  $\text{CF}_4$  and  $\text{C}_2\text{F}_6$  in favor of e.g.  $\text{CF}_3\text{H}$  but, above all, formation of solid  $\text{CaF}_2$  and  $\text{Ca}_5(\text{PO}_4)_3\text{F}$ .
  - Most studies pointed to thermal treatment as an effective solution for PFAS waste



Source: pfasfacts.com

# Solids Management Options



Source: GWI

# Thermal Treatment of Sludge

New and old technologies addressing the PFAS issue



| Incineration                      | Supercritical Water  | Pyrolysis/Gasification  | BioCon ERS Furnace           |
|-----------------------------------|----------------------|-------------------------|------------------------------|
| <u>900</u> – 1,600 F              | 705 – 1,100 F        | <u>450</u> – 1,650 F    | 800 – 2,100 F                |
| < 6.5 sec (FB), > 10 min (MH) SRT | < 1 min SRT          | > 75 min SRT            | Adjustable                   |
| < 6.5 sec (FB), Varies (MH) gas   | ?                    | <u>Syngas combusted</u> | Adjustable                   |
| <u>+25% DS</u>                    | +15% DS              | +80% DS                 | +90% DS                      |
| Dewatered Cake                    | Dewatered Cake       | Dried solids            | Dried solids                 |
| <b>Ash</b> water, wet gas         | <b>Water</b> wet gas | <b>Char</b> water, gas  | <b>Ash</b> <u>water, gas</u> |
| Proven                            | <u>Emerging</u>      | Emerging                | Proven                       |



# What do we know?

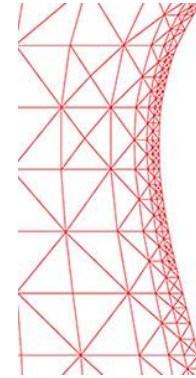
- Partial PFAS destruction is possible using thermal treatment above 600° C
  - How do we get complete PFAS destruction via TO?
  - How to quantify PFAS destruction rates in solids, liquids and gases?

# Research



# Fate of PFAS in Sludge & Emission

- Laboratory scale test
- Self-funded study
- Phase I: PFAS transport test
- Phase II: Thermal decomposition test
- Phase III: Site specific testing



University of Dayton  
Research Institute

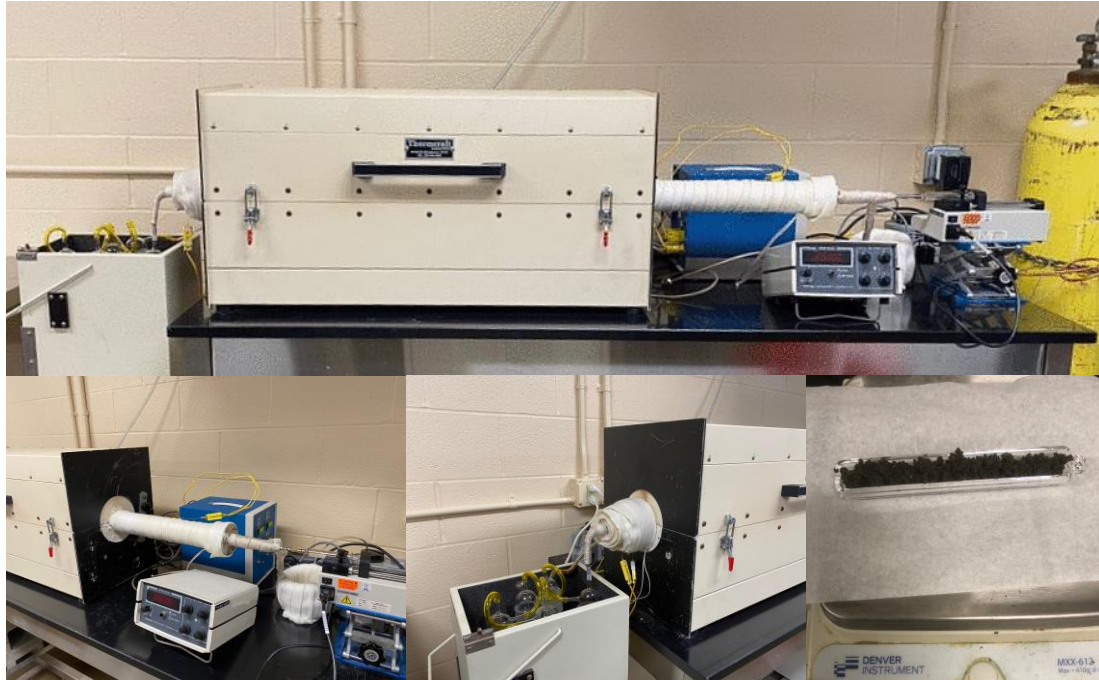


# Study Deliverables

- Verify thermal conditions for destruction of PFAS
  - Develop a Fluoride mass balance
- Identify chemical pre-treatment to reduce regulated emission
  - Acquire knowledge to address this emerging market with data driven design/solutions
- Determine design criteria to fine tune
  - ERS for PFAS destruction
  - APC for emission



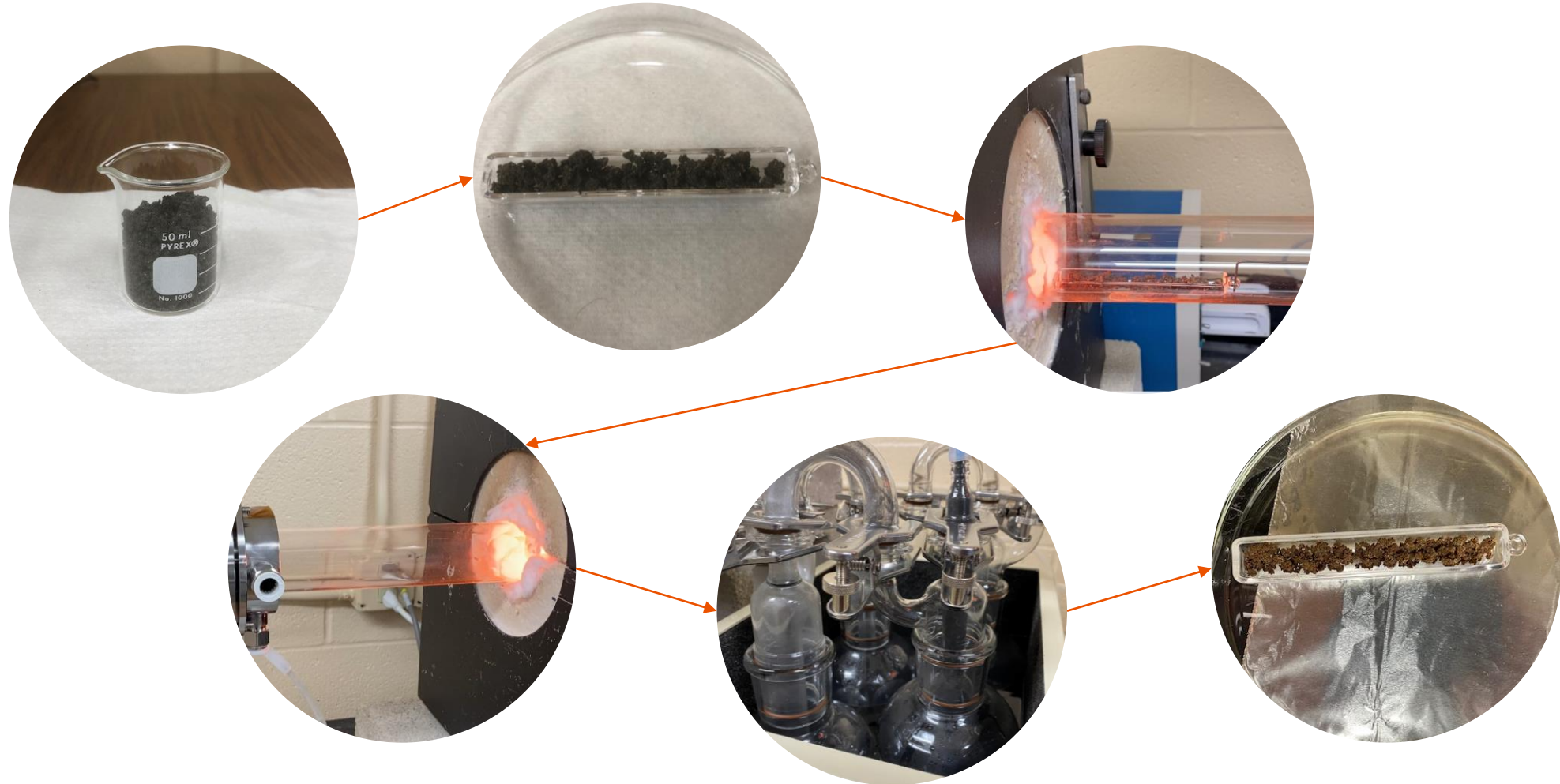
# Laboratory Scale Simulation of ERS



- Phase I
  - GC-MS was not connected to the reactor system
  - No Continuous Gas Analyzer (CGA)
    - Instead we used another CO monitor to check if the incineration goes under stoichiometry.
    - CO is a good indicator to check if enough oxygen is supplied or not.
- Phase II
  - GC-MS to monitor O<sub>2</sub>, CO<sub>2</sub>, N<sub>2</sub>, and H<sub>2</sub>O.
  - Supplied gas is only air and N<sub>2</sub> and no cal. Gas.
- Phase III
  - Site specific testing
  - Complete HF balance
  - Spiking if/when needed



# Solids, liquids and gases



# Challenges

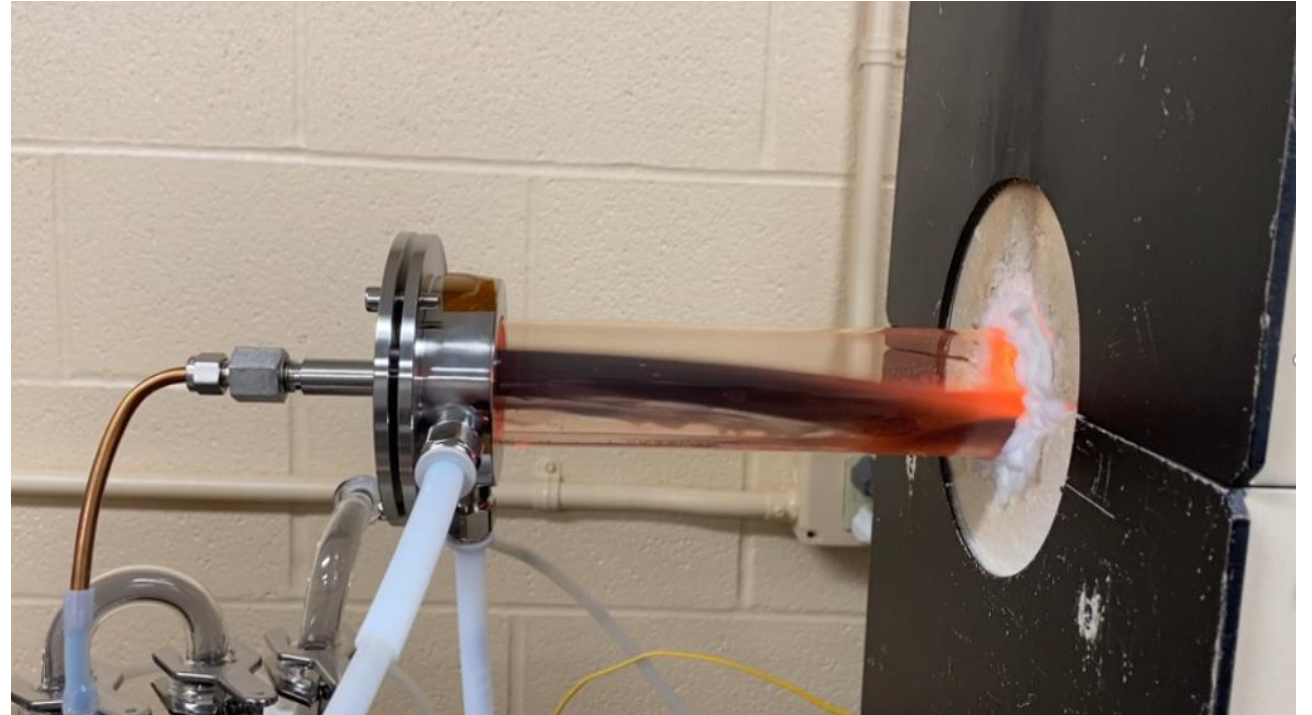
- Sites with high concentration of PFAS in sludge (Phase I and II)
  - Extensive search and numerous NDA
  - Sample locations redacted and erased
  - Samples acquired and dried to +90% DS using Veolia laboratory set up to simulate BioCon dryer
- Sample volume vs. ash
  - Limited sample boat size
  - Repetition of test beyond test requirement so collect ash



Total weight = 325 g  
Tray weight = 208 g > 117 g  
w/fin Bottle = 17 g  
Net Weight = 51 g > Sample wt  
α #1 27 g > 64 g to  
weight 91 g

# Challenges

- Combustion air vs. emission capture
  - Air greater than stoichiometry
  - Increasing impinger volume increased capture efficiency
  - Soot formation
- Fluoride balance and need for steam cleaning
  - Flood the reactors and interconnecting tubes to capture any PFAS bound to walls of reactor/tubing



# Progress to date

## Method: 537 (modified) - Fluorinated Alkyl Substances

| Analyte  | Result | Qualifier | RL  | MDL  | Unit  |
|--|--------|-----------|-----|------|-------|
| Perfluorobutanoic acid (PFBA)                            | 1.7    | JB        | 9.0 | 1.3  | ug/Kg |
| Perfluoropentanoic acid (PFPeA)                          | ND     | F1        | 9.0 | 3.5  | ug/Kg |
| Perfluorohexanoic acid (PFHxA)                           | 8.8    | J         | 9.0 | 1.9  | ug/Kg |
| Perfluoroheptanoic acid (PFHpA)                          | ND     |           | 9.0 | 1.3  | ug/Kg |
| Perfluorooctanoic acid (PFOA)                            | ND     | F1        | 9.0 | 3.9  | ug/Kg |
| Perfluorononanoic acid (PFNA)                            | 2.0    | J         | 9.0 | 1.6  | ug/Kg |
| Perfluorodecanoic acid (PFDA)                            | 6.3    | J F1      | 9.0 | 0.99 | ug/Kg |
| Perfluoroundecanoic acid (PFUnA)                         | ND     | F1        | 9.0 | 1.6  | ug/Kg |
| Perfluorododecanoic acid (PFDoA)                         | ND     | F1        | 9.0 | 3.0  | ug/Kg |
| Perfluorotridecanoic acid (PFTriA)                       | ND     | F1        | 9.0 | 2.3  | ug/Kg |
| Perfluorotetradecanoic acid (PFTeA)                      | ND     |           | 9.0 | 2.4  | ug/Kg |
| Perfluorobutanesulfonic acid (PFBS)                      | ND     |           | 9.0 | 1.1  | ug/Kg |
| Perfluoropentanesulfonic acid (PFPeS)                    | ND     |           | 9.0 | 0.90 | ug/Kg |
| Perfluorohexanesulfonic acid (PFHxS)                     | 3.1    | J I F1    | 9.0 | 1.4  | ug/Kg |
| Perfluoroheptanesulfonic Acid (PFHpS)                    | ND     |           | 9.0 | 1.6  | ug/Kg |
| Perfluorooctanesulfonic acid (PFOS)                      | 87     |           | 23  | 9.0  | ug/Kg |
| Perfluoronanesulfonic acid (PFNS)                        | ND     |           | 9.0 | 0.90 | ug/Kg |
| Perfluorodecanesulfonic acid (PFDS)                      | ND     | F1 F2     | 9.0 | 1.8  | ug/Kg |
| Perfluorooctanesulfonamide (FOSA)                        | ND     | F1        | 9.0 | 3.7  | ug/Kg |
| N-methylperfluorooctanesulfonamidoacetic acid (NMeFOSAA) | ND     |           | 90  | 18   | ug/Kg |
| N-ethylperfluorooctanesulfonamidoacetic acid (NEtFOSAA)  | ND     |           | 90  | 17   | ug/Kg |
| 4:2 FTS  | ND     |           | 90  | 17   | ug/Kg |
| 6:2 FTS  | 12     | J F1      | 90  | 6.8  | ug/Kg |
| 8:2 FTS  | ND     |           | 90  | 11   | ug/Kg |

## ● Phase I

- Five TE studies completed at various design conditions
- Emission capture efficiency being fine tuned

## ● Phase II

- PFAS volatilization and destruction rates at varying
  - Concentrations
  - Temperature
  - O2 concentration
  - Dwell times - solids and gases
- TOP Assay to round of F balance

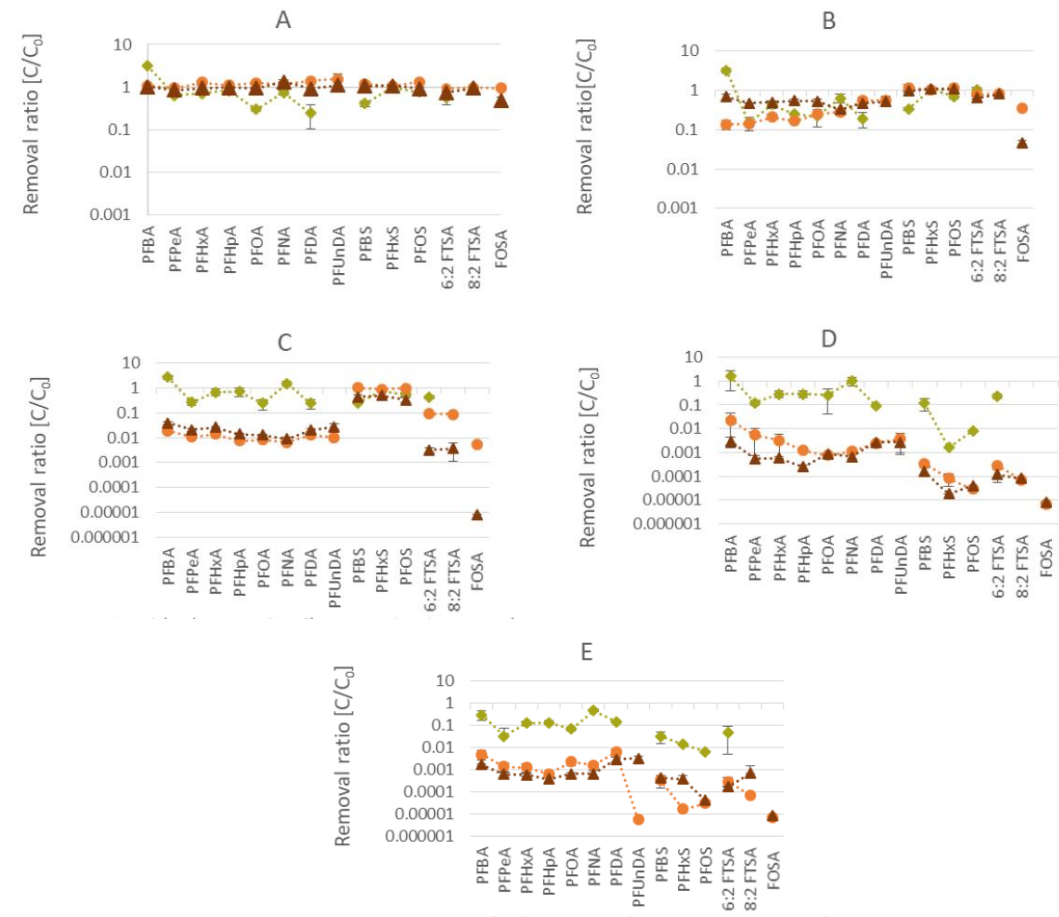
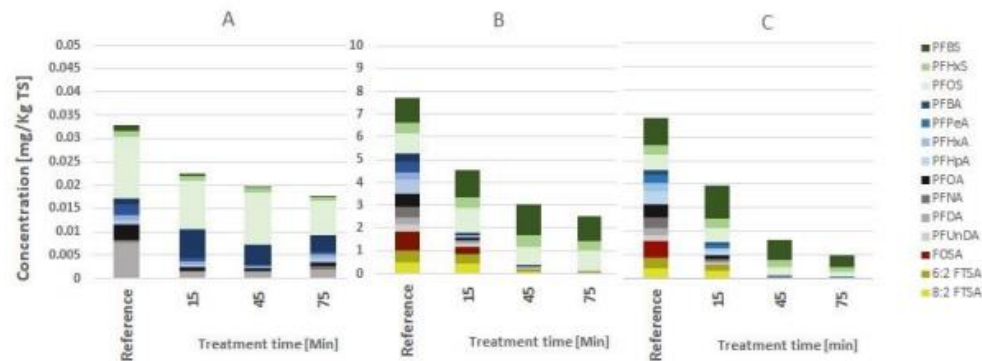
## ● Phase III

- In progress
  - Paid feasibility study opportunity

# Early Results

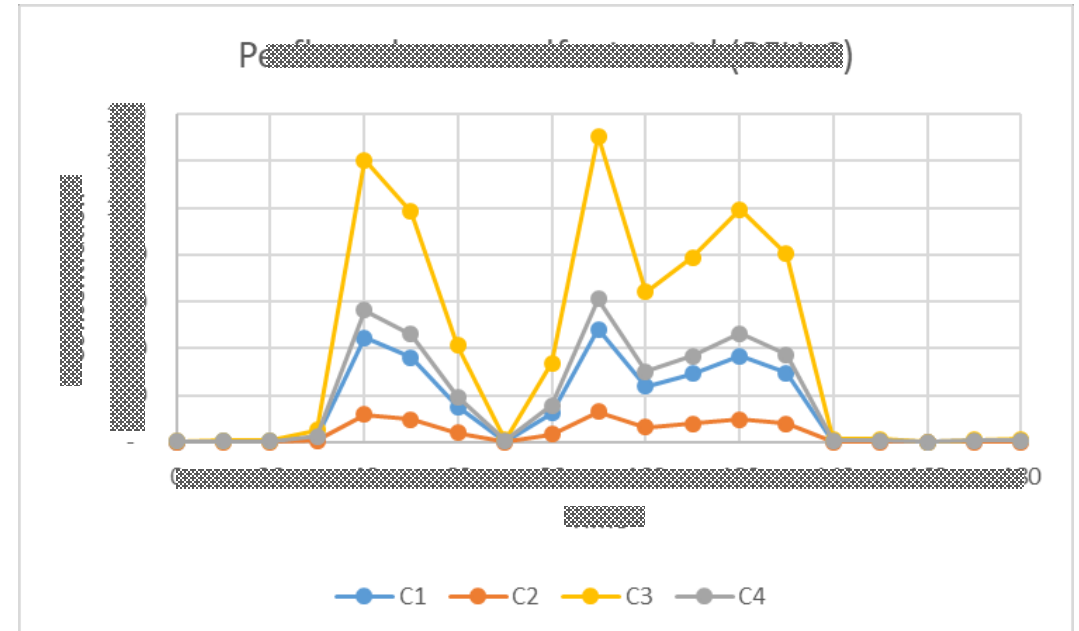
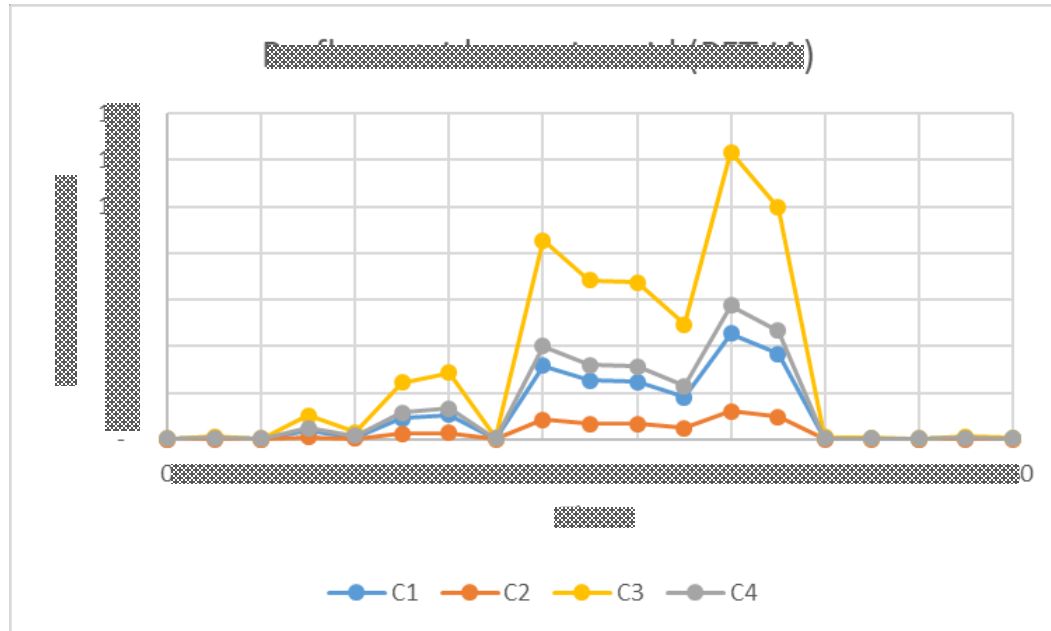
- Verifying Literature Values on Bench Scale

- Volatilization of PFAS compounds subjected to various thermal oxidative conditions
- Determine the vapor pressure and desorption potential for each compounds
- Data used to fine tune dryer model



Source: An assessment of thermal desorption as a remediation technique for per- and polyfluoroalkyl substances (PFASs) in contaminated soil - Anna-Stina Lind

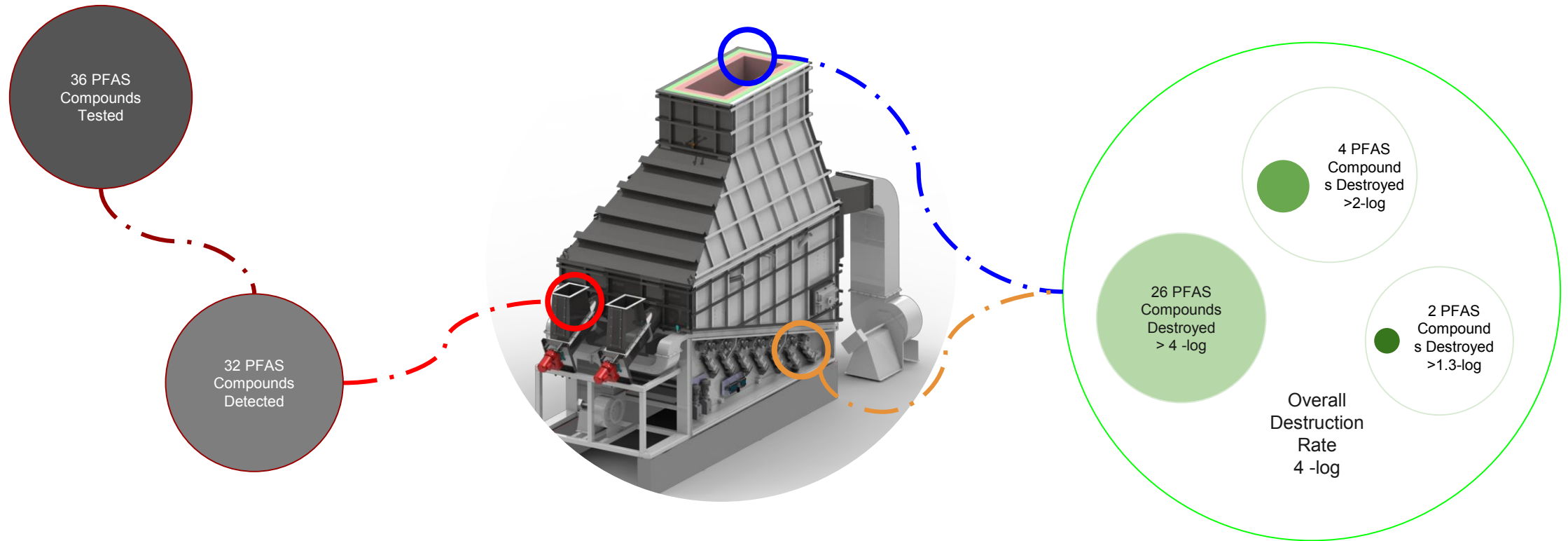
# Modeling



- Calibration

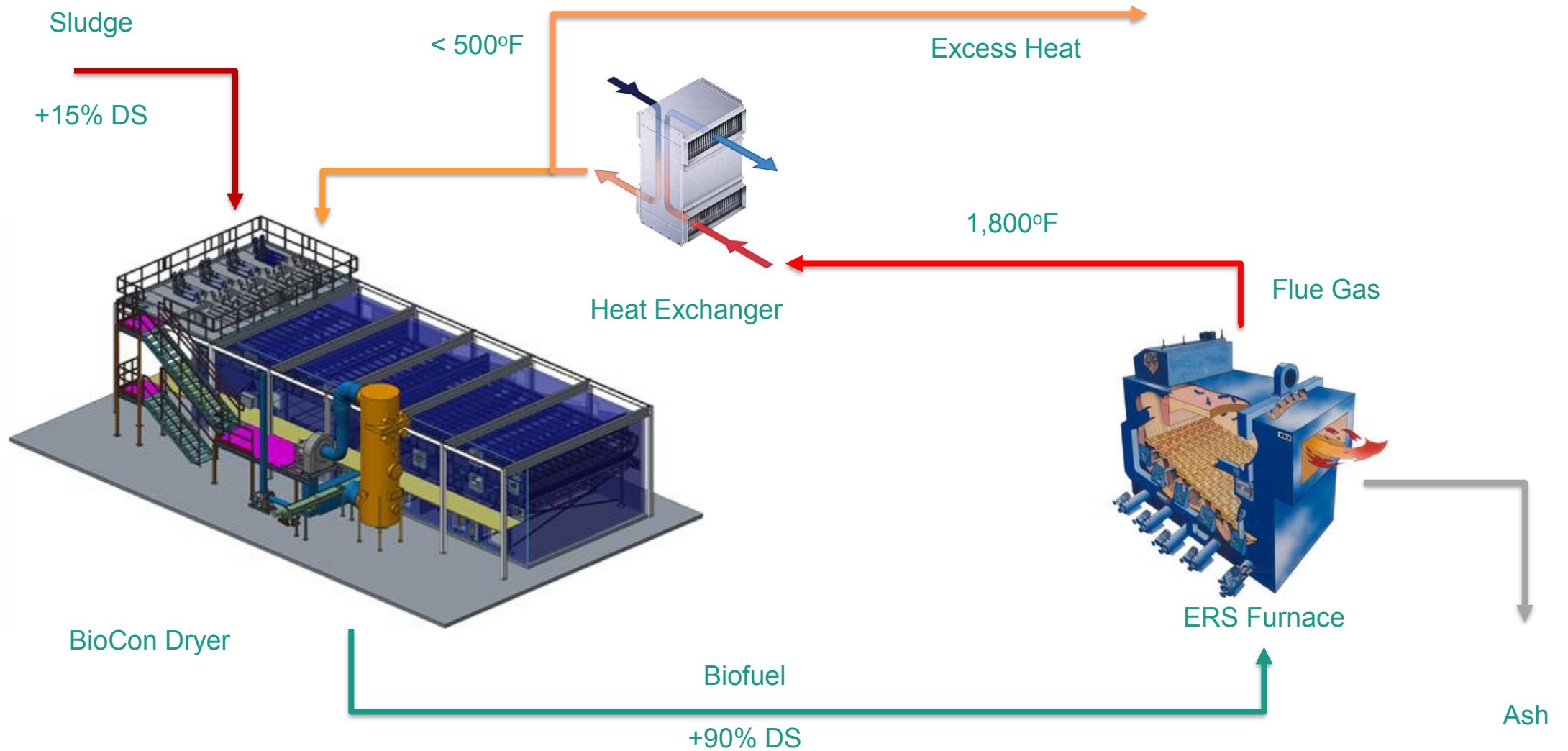
- T99 identified for various PFAS compounds
- Determine the minimum temperature for each compounds
- Data used to fine tune ERS model, update APC efficiencies

# Oxidation of PFAS Laden Sludge



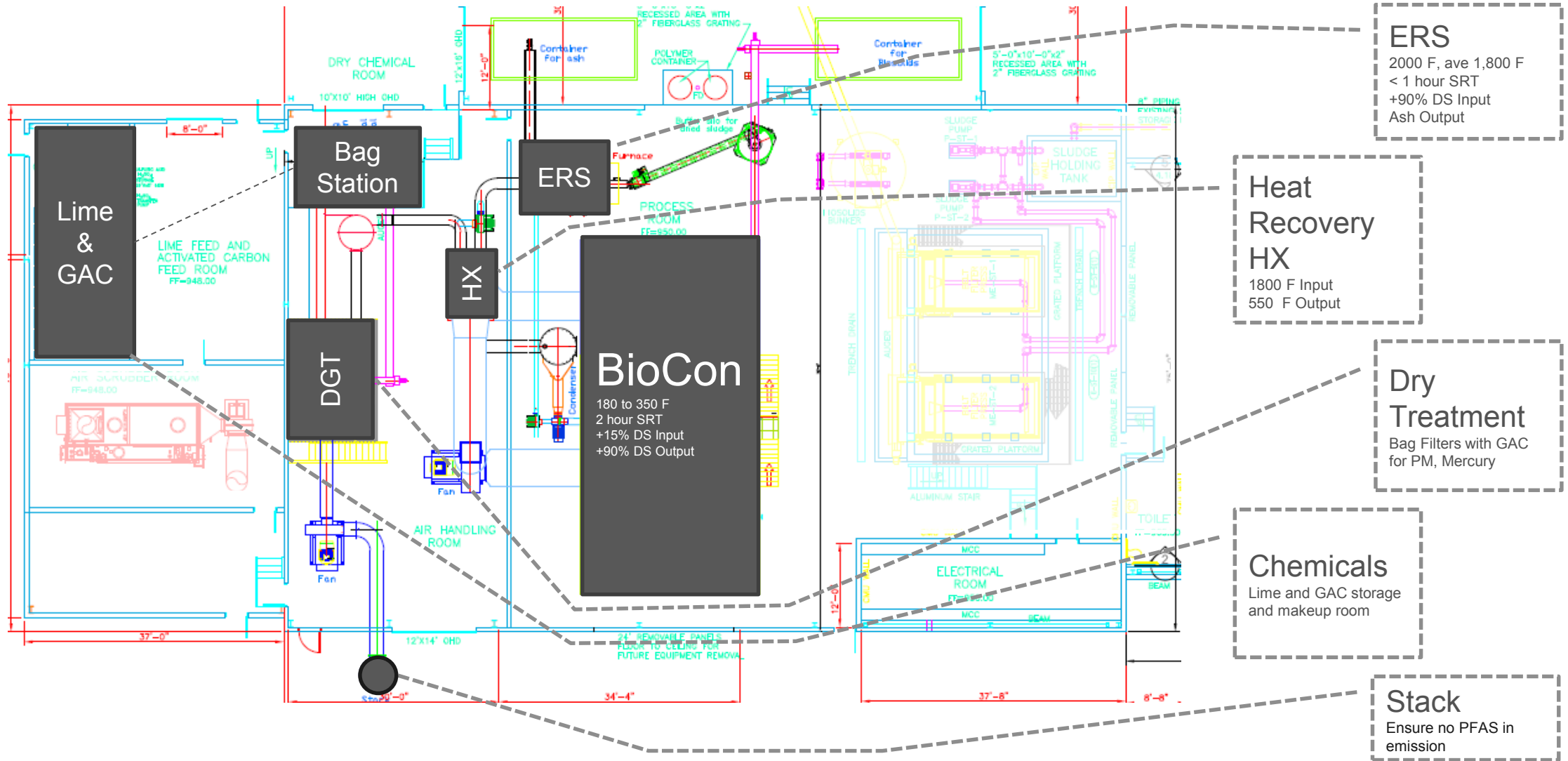
- PFAS laden sludge input
- Ash output
- Emissions output

# BioCon ERS – Highlights

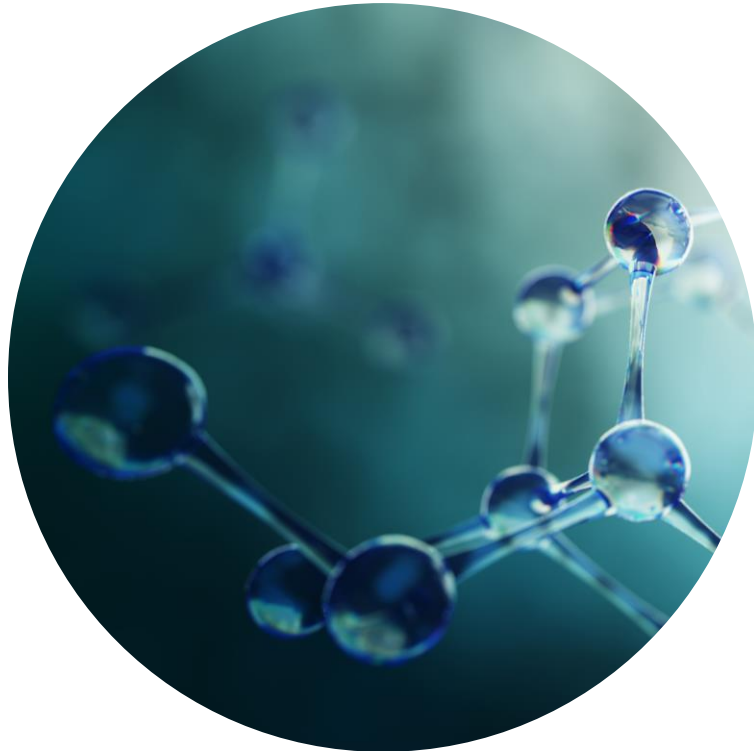




# Experience in the US (Buffalo, MN - 2008)



# Summary



## Is complete destruction of PFAS in sludge possible?

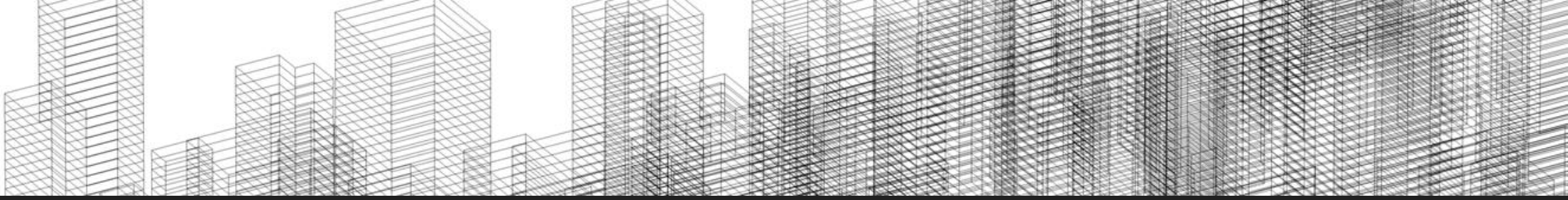
- ERS and associated APC, at optimized operating conditions, can effectively destroy and remove PFAS from sludge, process liquids and emission
- Longer the chemical chain, the higher the decomposition efficiency
- Short chain compounds can form due to incomplete RT

## Other findings

- Emission data is promising, APC impacts project cost
- HF acid management should be integral to the emission control train
- F balance is difficult, but possible with right techniques

## Unknowns

- Fate of PFAS subjected to 'incomplete' or 'low temperature' combustion requires further study, example, PFAS in emissions



# Thank you

For PFAS destruction in solids, contact:

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Contact our local representatives:

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[dcoppola@frmahony.com](mailto:dcoppola@frmahony.com)

