



Environmental Sciences Research Institute

# **Treatment of Wastewater from Steel Industry Using Various Type of Natural and Chemical Coagulants**

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Steel industries are major contributors to the world economy, especially in developing nations

Overview

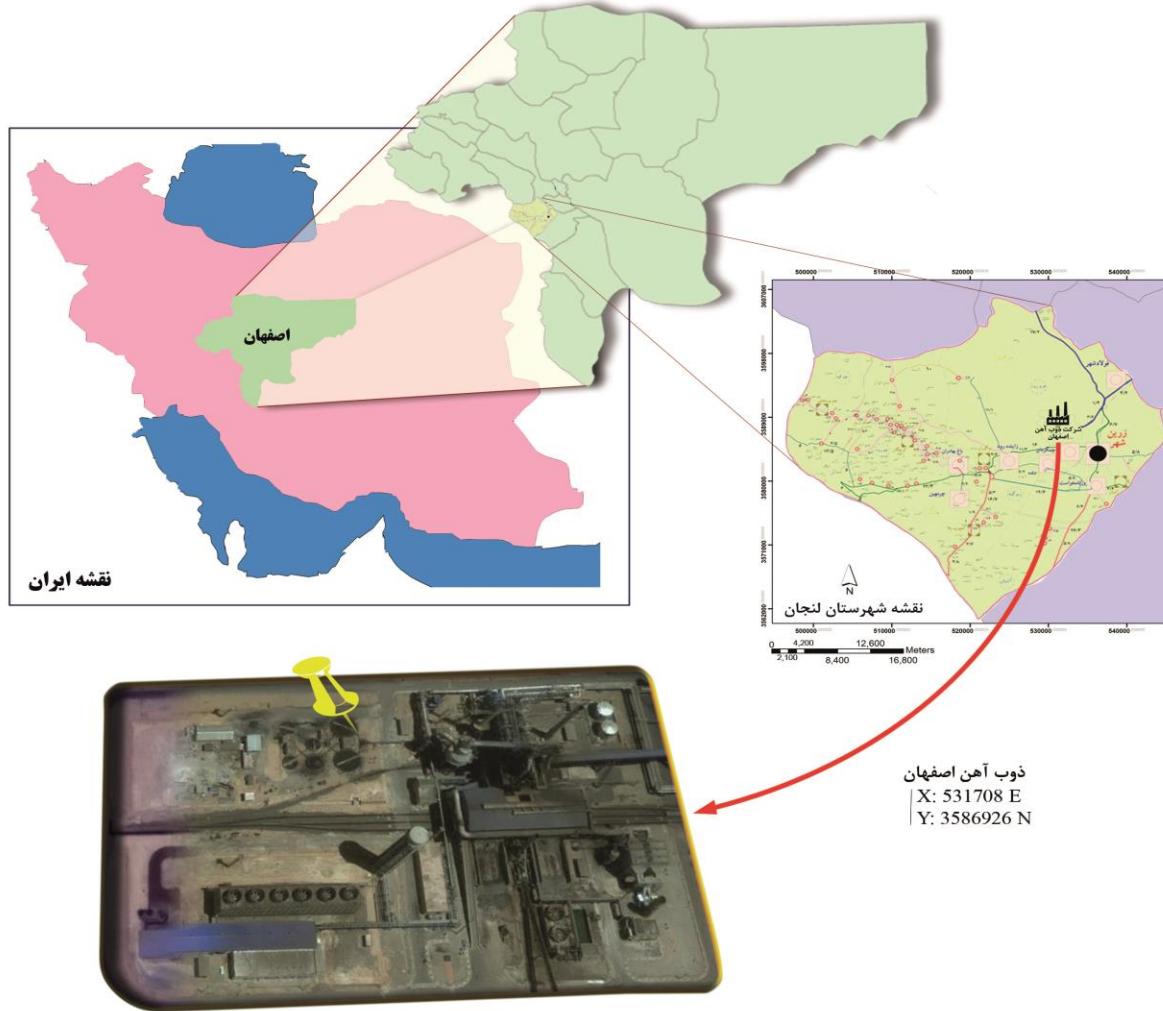
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Steel industries are one of the main contributors to industrial pollution worldwide



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# Natural and Chemical Coagulants

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**Ferric Chloride**  
 $(\text{FeCl}_3 - 6\text{H}_2\text{O})$

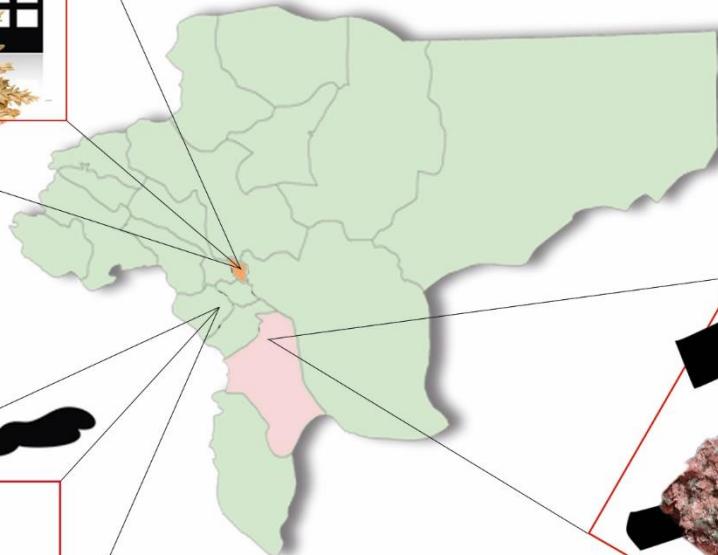
**Alum**  $[\text{Al}_2(\text{SO}_4)_3 \cdot 16\text{H}_2\text{O}]$

**Poly Aluminum Chloride (PAC)**  
 $[\text{Al}_2(\text{OH})_n\text{Cl}_6 - n \cdot x\text{H}_2\text{O}]_m$

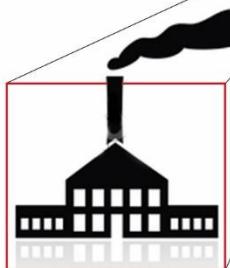
**Dolomite Powder**

**Wheat Starch**

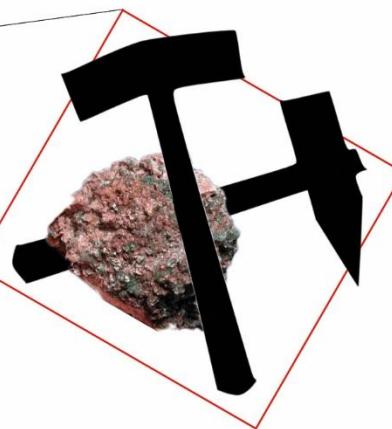
## Amin Starch Manufacturer



## Esfahan Steel Company



## Shahreza Dolomite Mine



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## Physico-chemical characteristics of starch coagulant

Description of test	Standard
Appearance	White
Smell and taste	Free of undesirable odors and flavors
pH	4.5-7
Humidity	Max 12 %
Purity	80-85 %
Total ash	Max 0.25 %
Proteins NX : 6/25	Max 0.4 %
Fat	Max 0.3 %
Physical Review	No foreign objects
Impurity	Wheat bran, sulfated ash and gluten
Type of abuse	Food Industry

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## Chemical analysis of dolomite

Compound Name	(%)Amount	Compound Name	(%)Amount
SO <sub>3</sub>	0.01 ≤ 0	CaO	31.69
P <sub>2</sub> O <sub>5</sub>	0.01 ≤ 0	MgO	20.6
K <sub>2</sub> O	0.01 ≤ 0	SiO <sub>2</sub>	0.4
Na <sub>2</sub> O	0.01 ≤ 0	Fe <sub>2</sub> O <sub>3</sub>	0.26
L.O.I	46.87	Al <sub>2</sub> O <sub>3</sub>	0.11



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## Physico-chemical characteristics of raw wastewater of blast furnace no.3

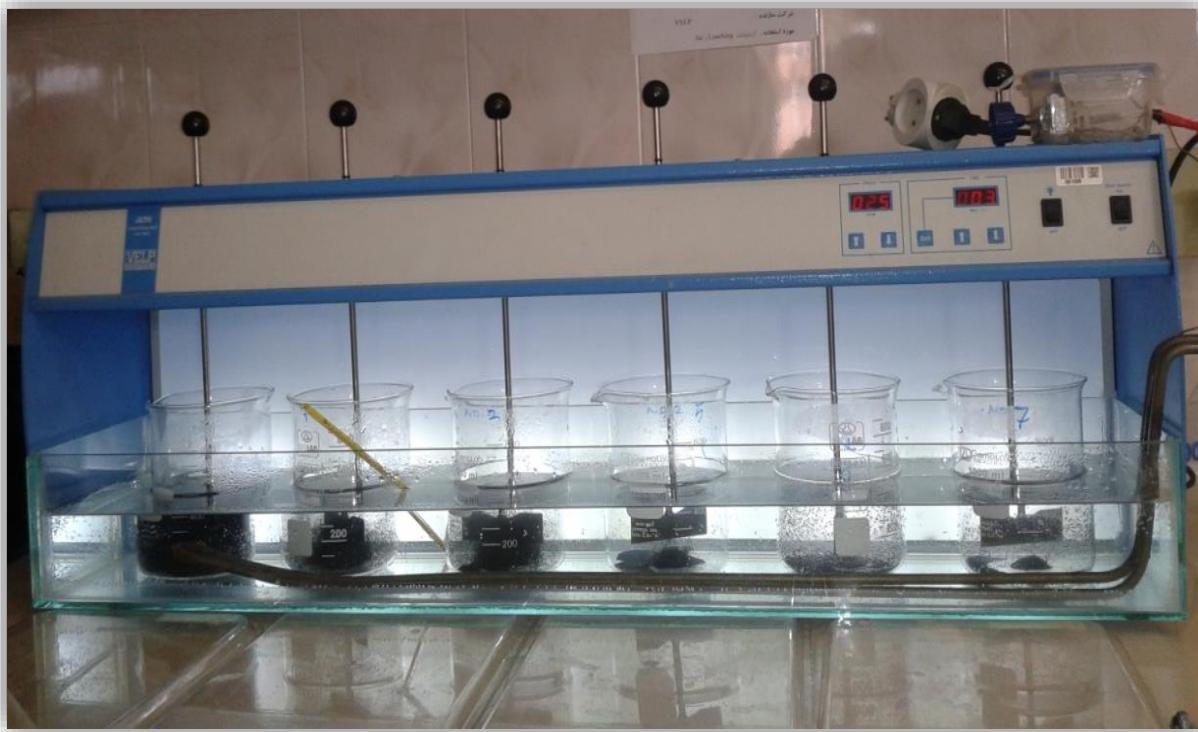
Parameters	Magnitude
pH	8.95
Temperature (C°)	50-55
Color	pitch-black
Odor	Coke
Chemical oxygen demand (mg l <sup>-1</sup> )	3750
Total dissolved solids (mg l <sup>-1</sup> )	17800
Total suspended solids (mg l <sup>-1</sup> )	9650
Total solids (mg l <sup>-1</sup> )	27450
EC (mhos/cm)	37.7
Turbidity (NTU)	1942
NH <sub>3</sub> (mg l <sup>-1</sup> )	NH <sub>3</sub> -N:435, NH <sub>3</sub> :530, NH <sub>4</sub> :560

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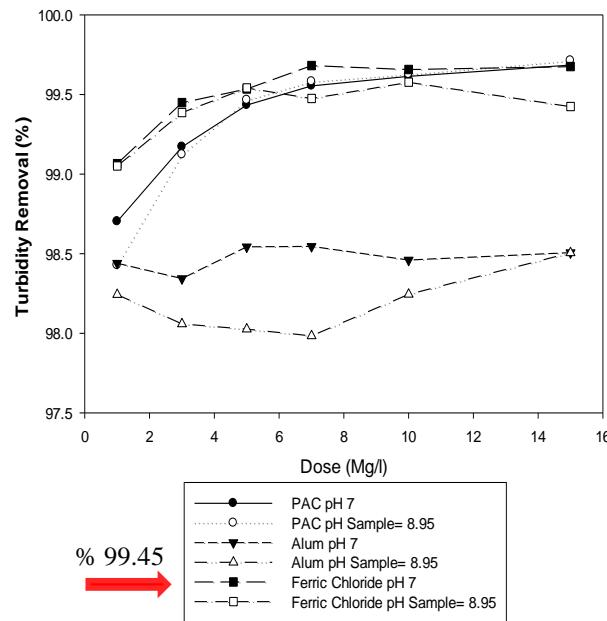
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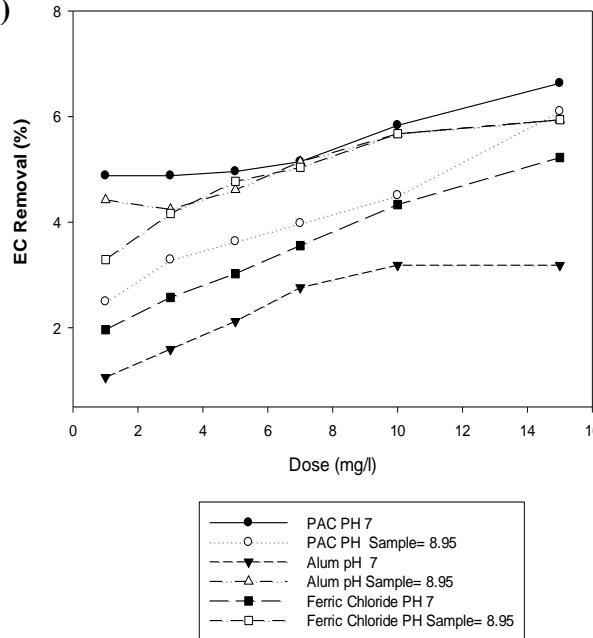
**Discussion**

Description	Factors
Natural pH of wastewater - 7	pH
PAC – Alum- $\text{FeCl}_3$ -Wheat Starch-Dolomite Powder	Coagulant Type
1 – 3 – 5 – 7 – 10 – 15 (ppm)	Coagulant dosage in Chemical Phase
5 – 10 – 30 – 60 – 100 – 150 (ppm)	Coagulant dosage in Natural Phase
0 – 2 – 5 – 10 – 30 – 60 (ppm) Natural Coagulants	Coagulant dosage in Mix Phase
1 – 1 – 1 – 1 – 1 – 1 (ppm) Chemical Coagulants	
Optimum Dose	

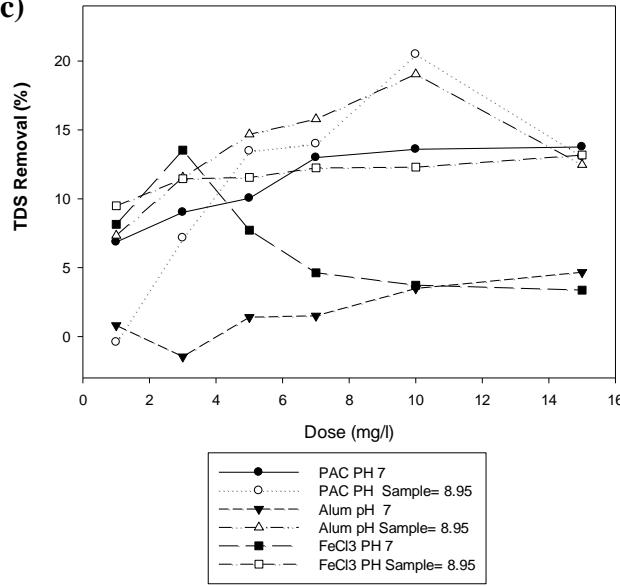
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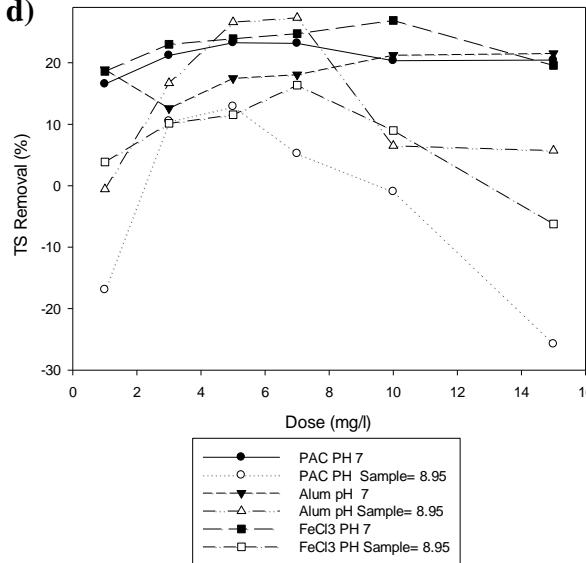
b)



c)



d)



## Overview

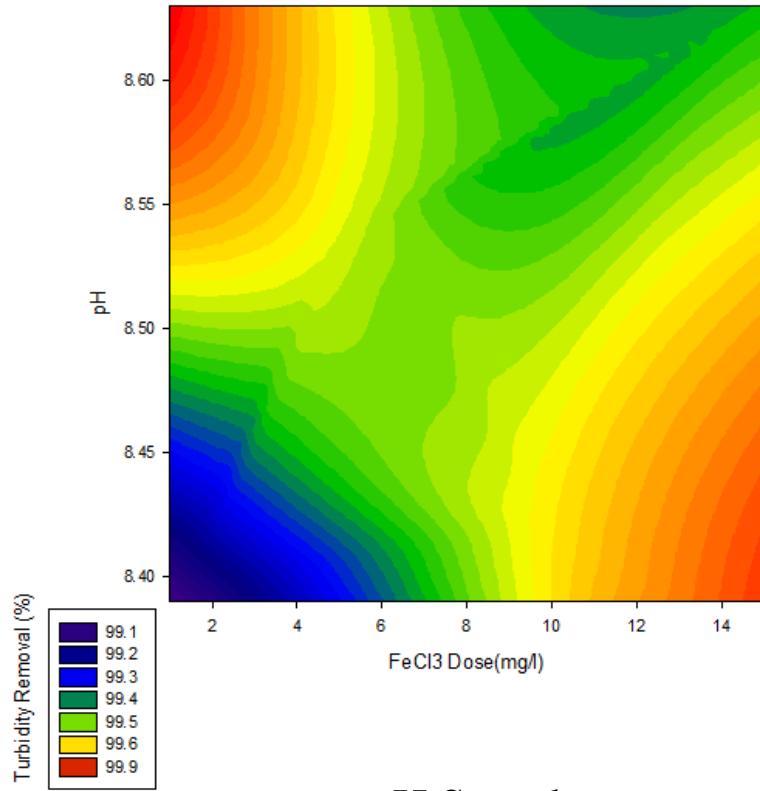
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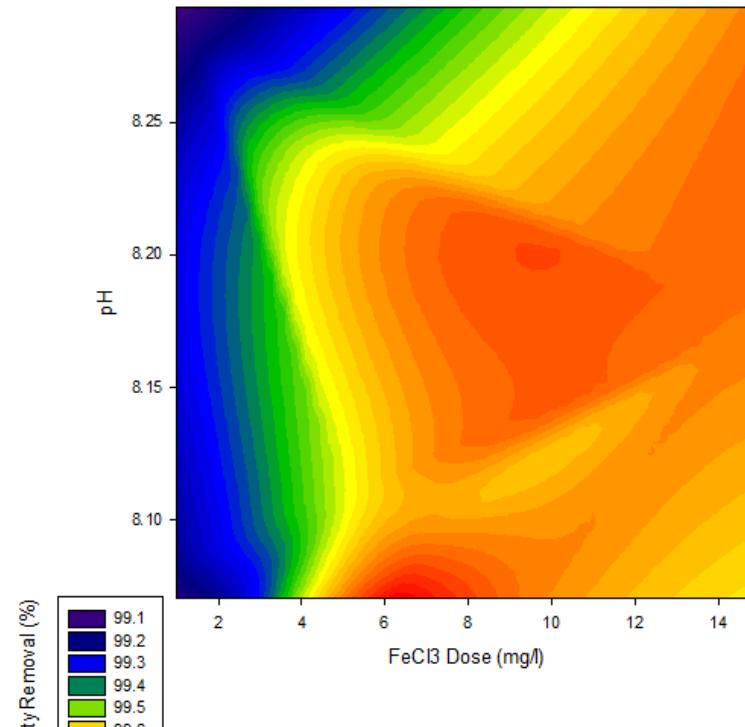
## Discussion

Effect of various chemical coagulant dosages (PAC, alum,  $\text{FeCl}_3$ ) on a) turbidity removal pH (natural-7) b) EC removal pH (natural-7) c) TDS removal pH (natural-7) d) TS removal pH (natural-7).

## Effect of various factors on turbidity removal



pH Sample



pH7

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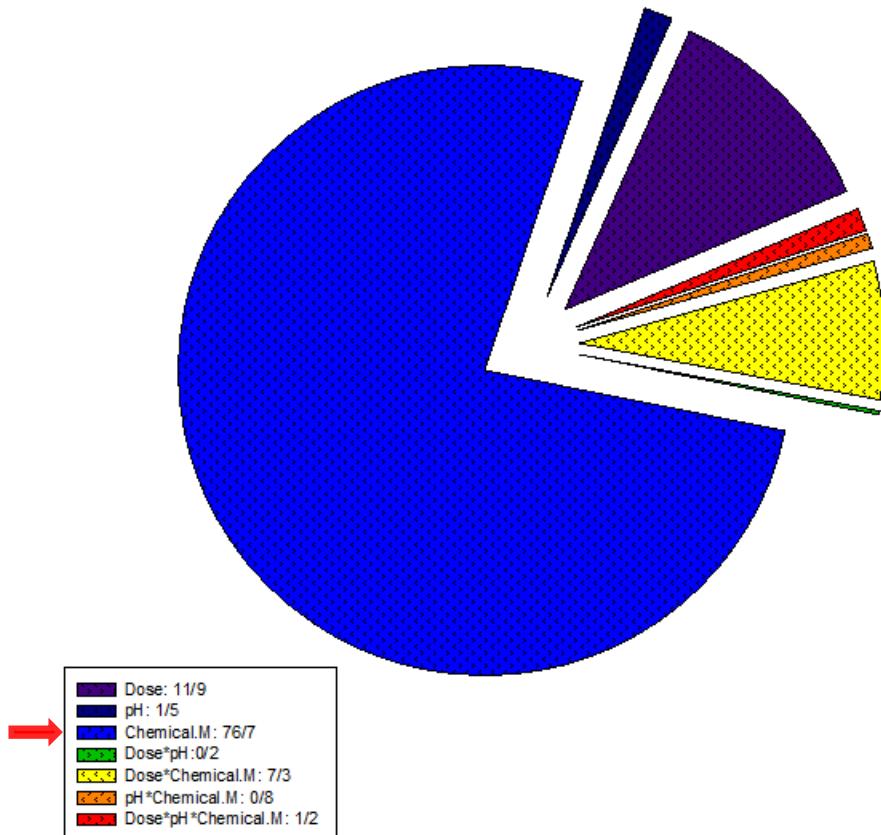
## Effect of various factors on turbidity removal

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Microscope Images of the flocs (mm) produced by 1 mg/l PAC, 1 mg/l alum and 1 mg/l  $\text{FeCl}_3$  at 30 min sedimentation time

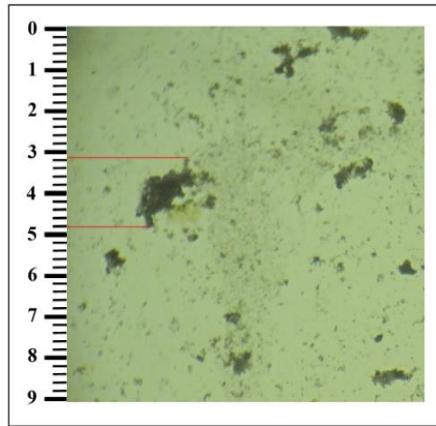
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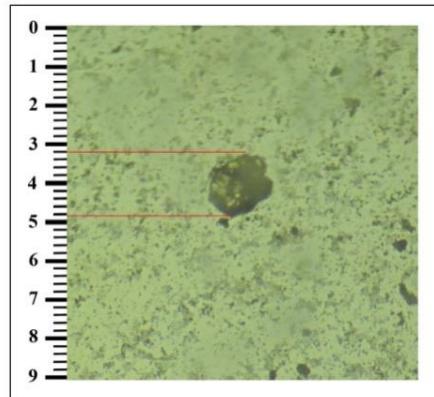
Results

Discussion

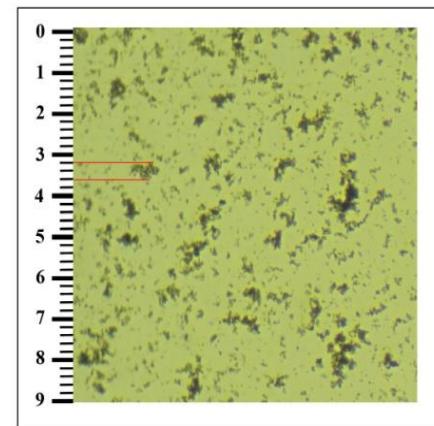
PAC



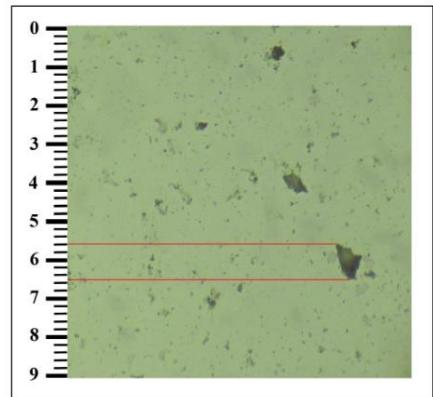
$\text{FeCl}_3$

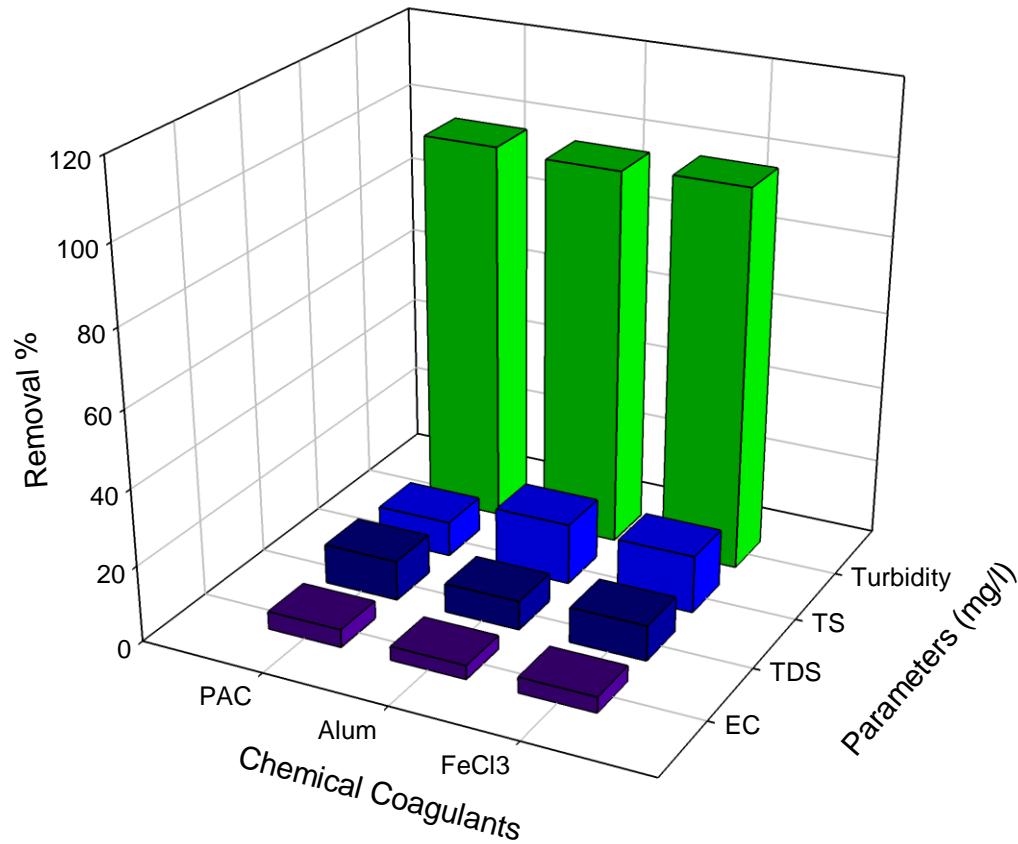


Raw Wastewater



Alum





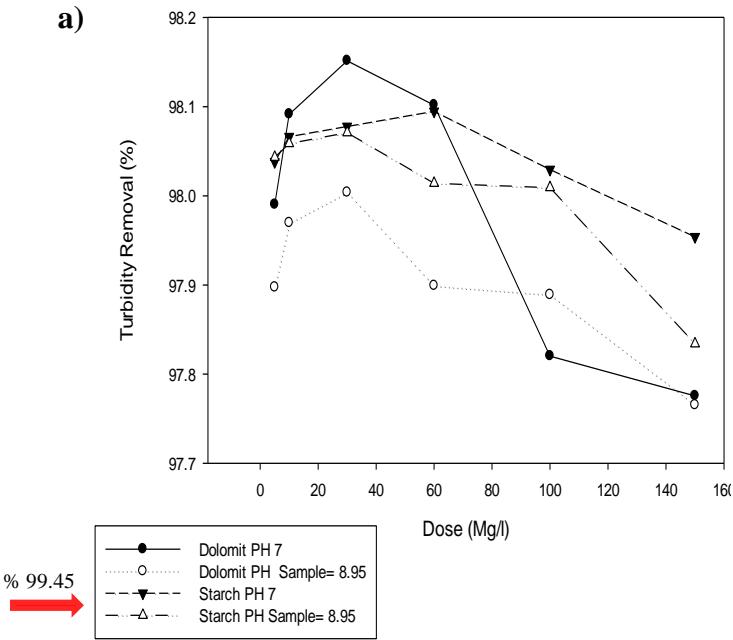
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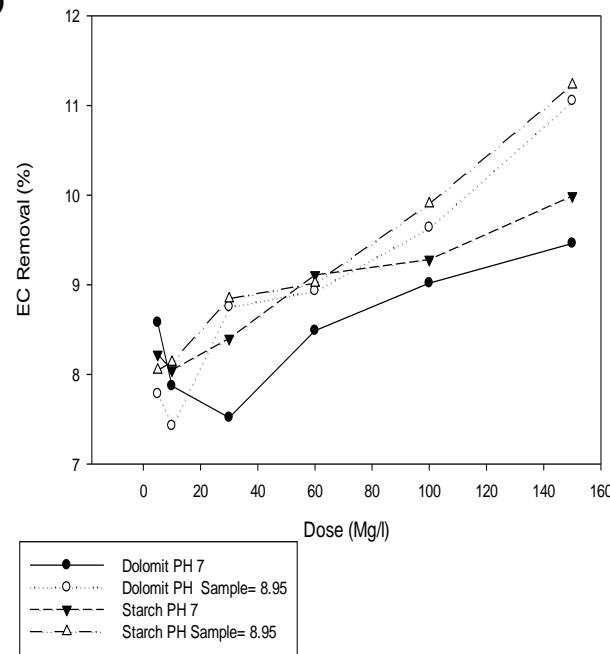
Results

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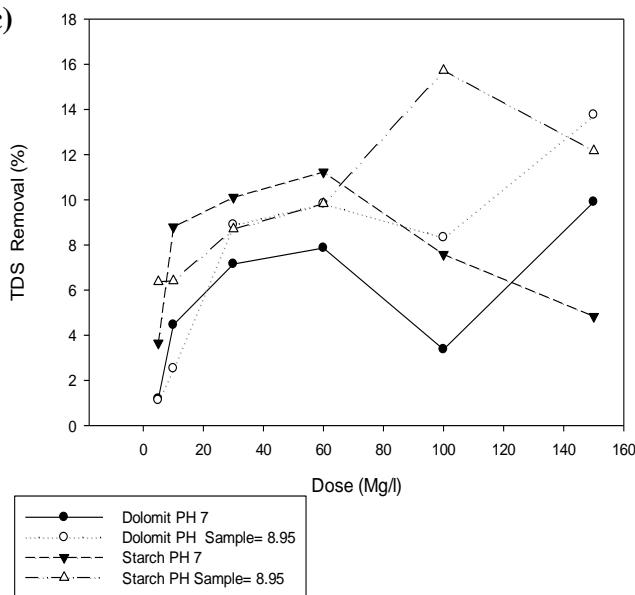
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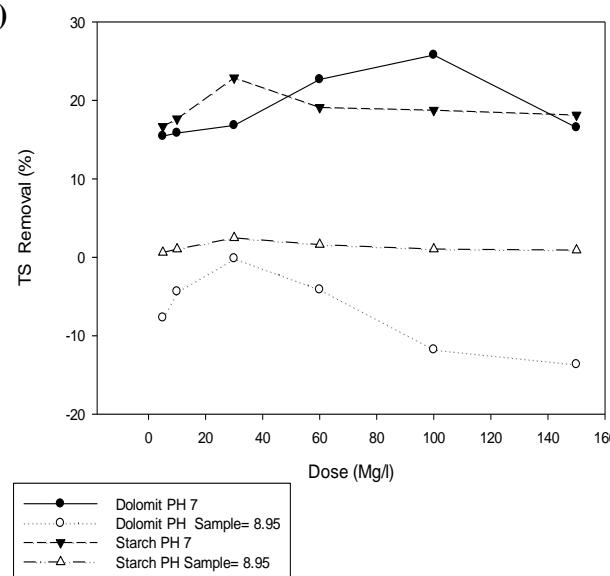
b)



c)



d)



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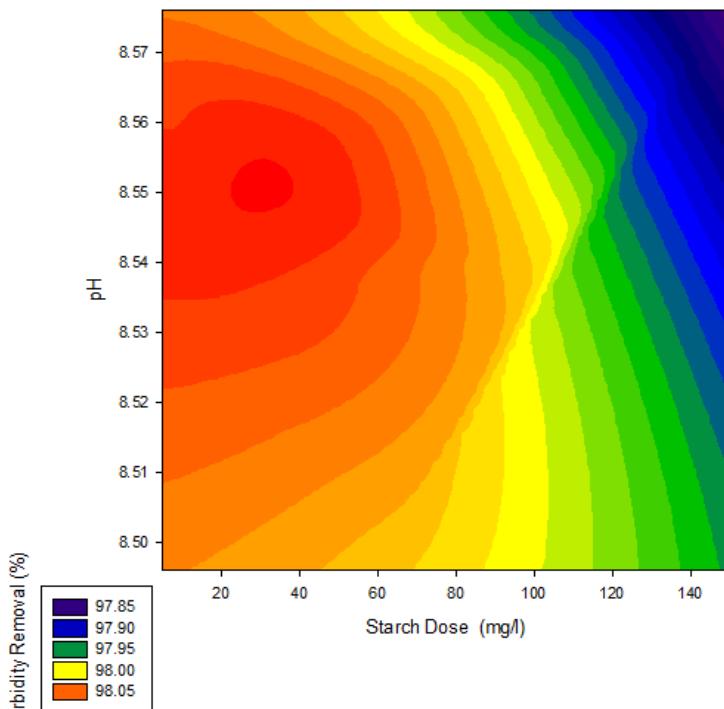
## Methodology

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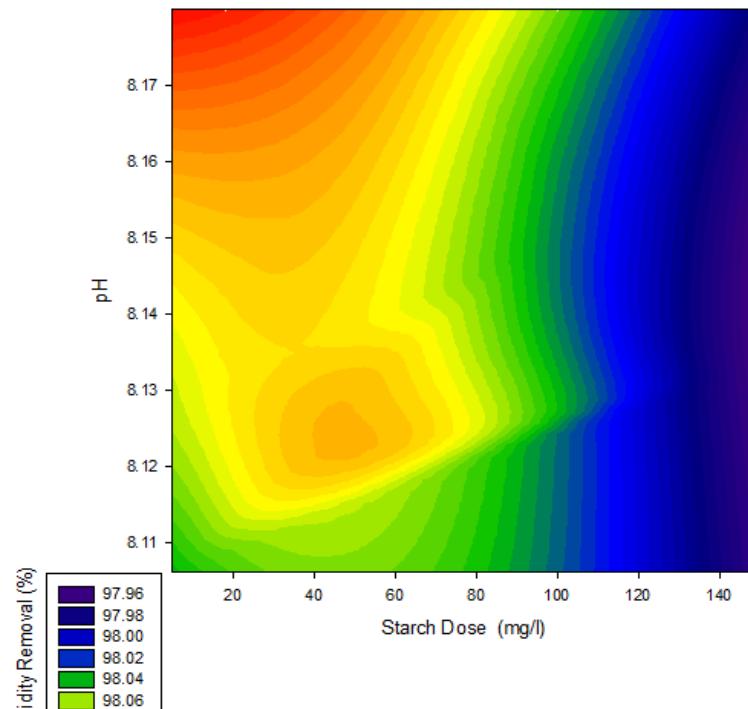
## Discussion

Effect of various natural coagulant dosages (wheat starch, dolomite) on  
a) turbidity removal pH (natural-7)  
b) EC removal pH (natural-7) c)  
TDS removal pH (natural-7) d) TS removal pH (natural-7).

## Effect of various factors on turbidity removal



pH Sample



pH7

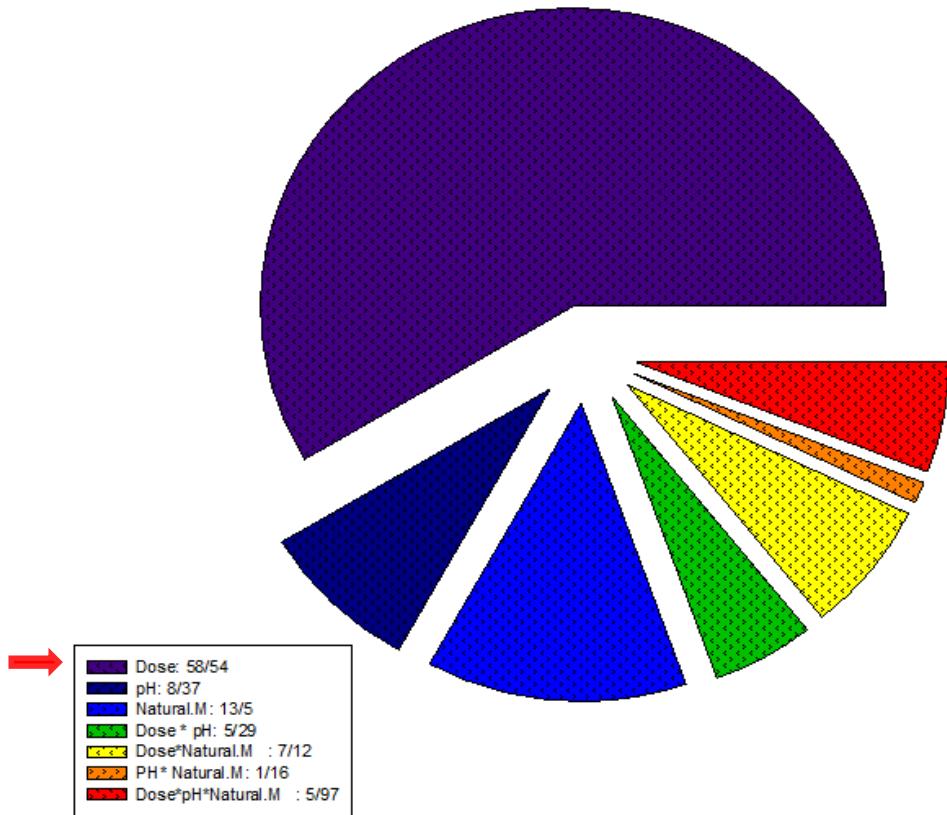
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## Effect of various factors on turbidity removal



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Microscope Images of the flocs (mm) produced by a) 5 mg/l dolomite and b) 5 mg/l wheat starch at 30 min sedimentation time

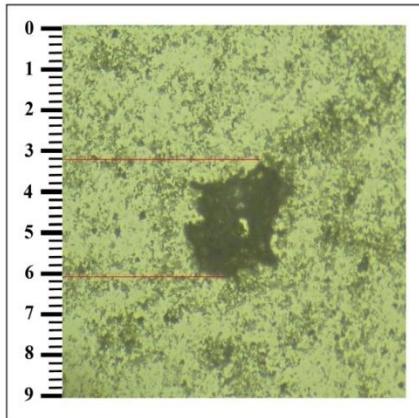
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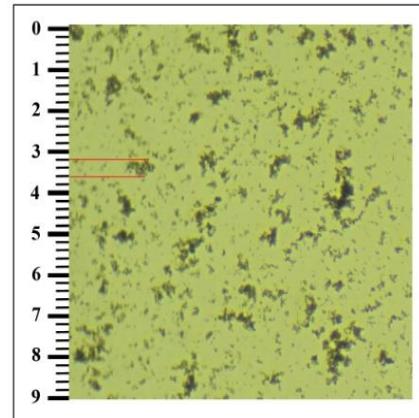
Results

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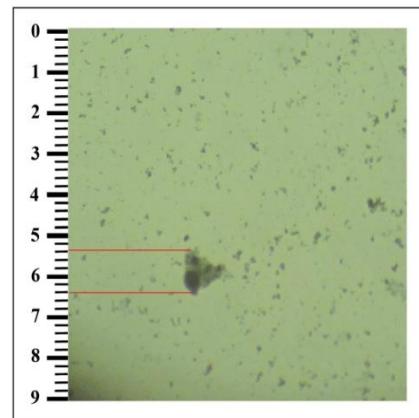
Dolomit

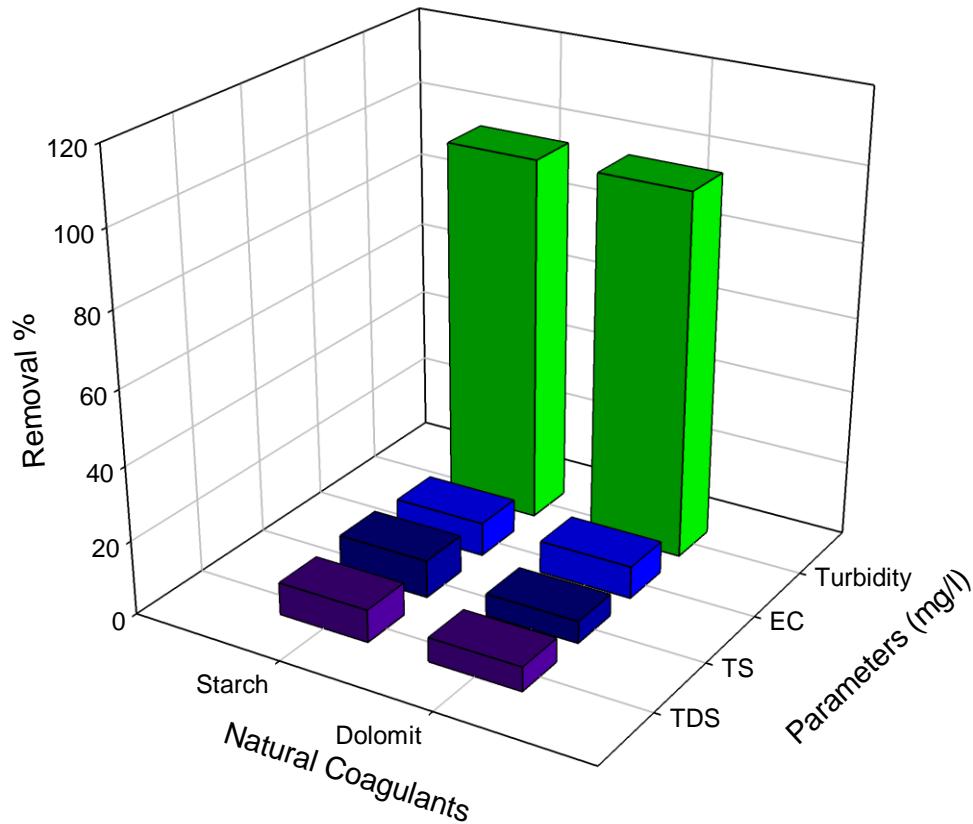


Raw Wastewater



Starch





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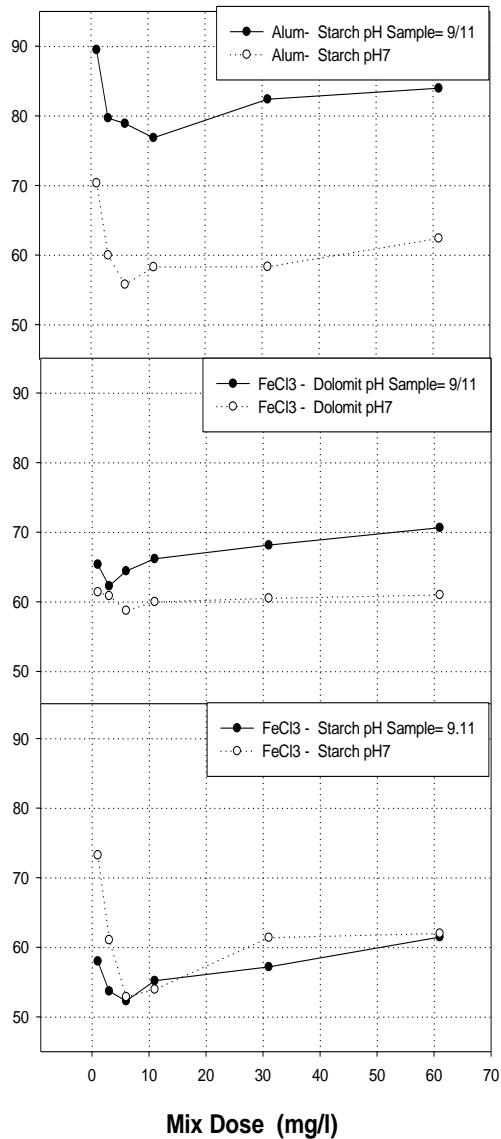
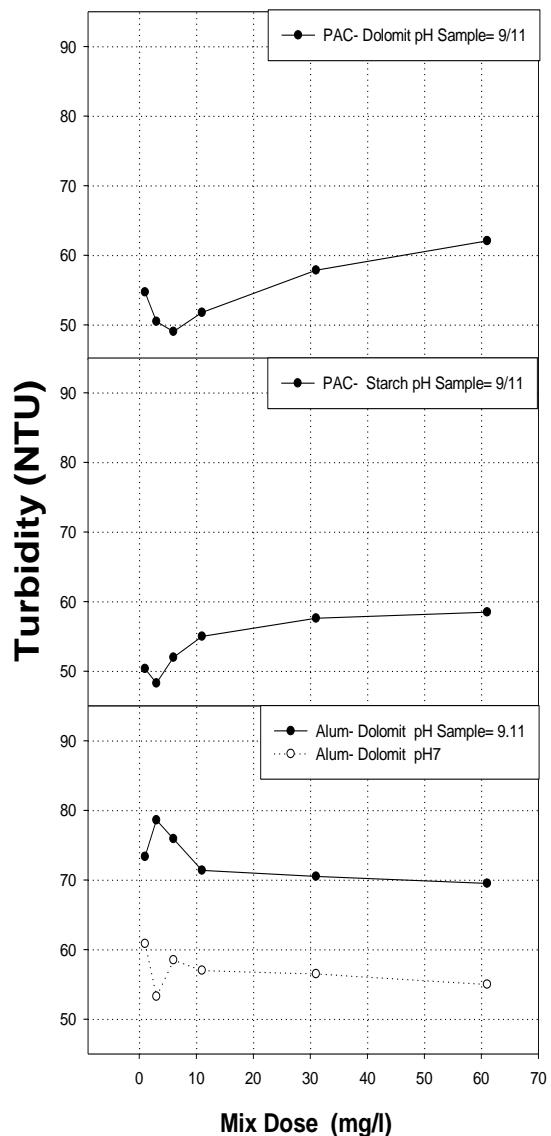
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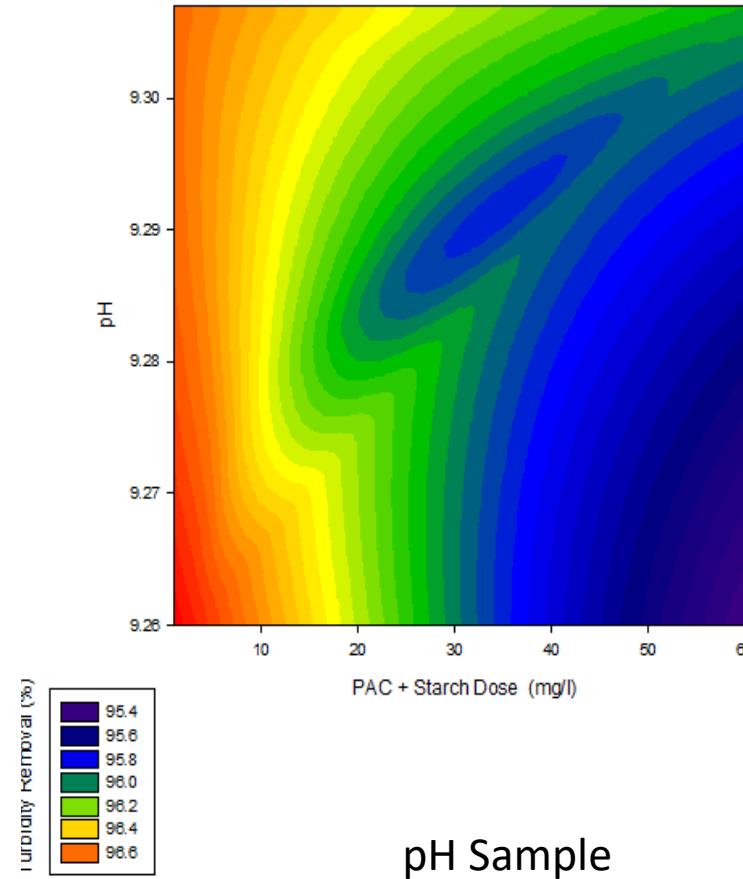
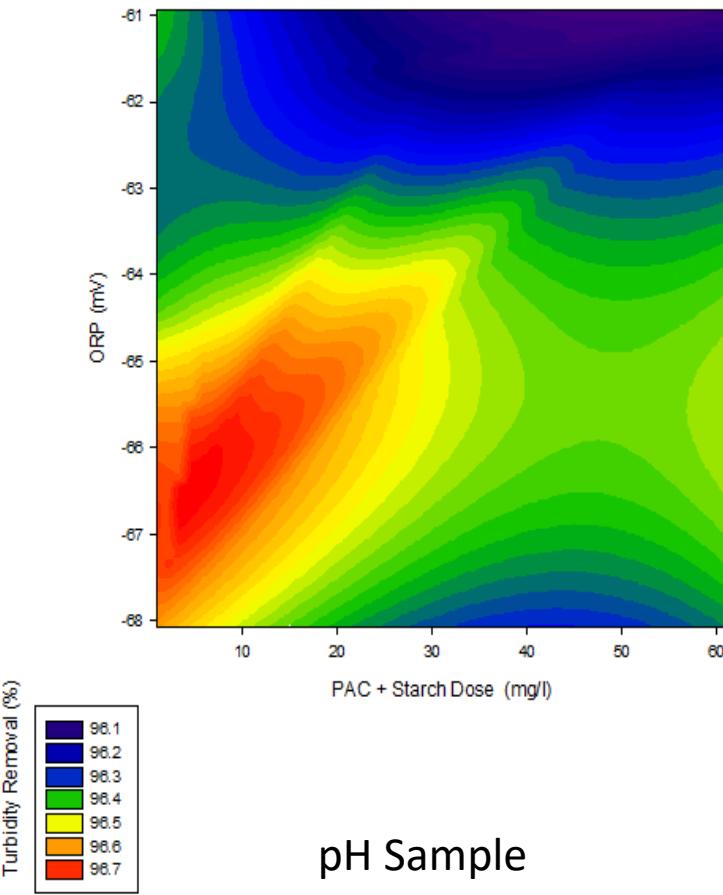
## Discussion



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Effect of chemical/natural coagulants combinations dosages on turbidity removal pH (natural-7)

## Effect of various factors on turbidity removal



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## Effect of various factors on turbidity and EC removal

Turbidity removal							EC removal						
Source	df	Sum of square	Mean square	F value	p value	Impact percent (p%)	Source	df	Sum of square	Mean square	F value	p value	Impact percent (p%)
X <sub>1</sub>	5	866.46	173.29	0.85	0.51	5.46	X <sub>1</sub>	5	47.81	9.56	3.54	0.0051	13.6
X <sub>2</sub>	1	884.36	884.36	4.33	0.03	5.58	X <sub>2</sub>	1	214.57	214.57	79.49	< 0.0001	61.05
X <sub>3</sub>	1	10.8	10.8	0.05	0.81	0.06	X <sub>3</sub>	1	0.16	0.16	0.06	0.80	0.04
X <sub>4</sub>	2	7623.96	3811.98	18.68	< 0.0001	48.11	X <sub>4</sub>	2	46.80	23.40	8.67	0.0003	13.31
X <sub>1</sub> X <sub>2</sub>	5	356	71.2	0.35	0.88	2	X <sub>1</sub> X <sub>2</sub>	5	3.91	0.78	0.29	0.91	1.11
X <sub>1</sub> X <sub>3</sub>	5	304.58	60.91	0.3	0.91	1.92	X <sub>1</sub> X <sub>3</sub>	5	4.61	0.92	0.34	0.88	1.31
X <sub>1</sub> X <sub>4</sub>	10	429.88	42.98	0.21	0.99	2.71	X <sub>1</sub> X <sub>4</sub>	10	3.03	0.3	0.11	0.99	0.86
X <sub>2</sub> X <sub>3</sub>	1	85.17	85.17	0.42	0.51	0.53	X <sub>2</sub> X <sub>3</sub>	1	3.79	3.79	1.40	0.23	1.07
X <sub>2</sub> X <sub>4</sub>	1	2928.61	2928.61	14.35	0.0002	18.48	X <sub>2</sub> X <sub>4</sub>	1	1.12	1.12	0.41	0.52	0.31
X <sub>3</sub> X <sub>4</sub>	2	1107.85	553.92	2.71	0.07	6.99	X <sub>3</sub> X <sub>4</sub>	2	0.02	0.01	0	0.99	0.005
X <sub>1</sub> X <sub>2</sub> X <sub>3</sub>	5	188.14	37.62	0.18	0.96	1.18	X <sub>1</sub> X <sub>2</sub> X <sub>3</sub>	5	4.23	0.84	0.31	0.90	1.2
X <sub>1</sub> X <sub>2</sub> X <sub>4</sub>	5	174.56	34.91	0.17	0.97	1.10	X <sub>1</sub> X <sub>2</sub> X <sub>4</sub>	5	3.98	0.79	0.3	0.91	1.13
X <sub>2</sub> X <sub>3</sub> X <sub>4</sub>	1	497.29	497.29	2.44	0.12	3.13	X <sub>2</sub> X <sub>3</sub> X <sub>4</sub>	1	3.77	3.77	1.4	0.23	1.07
X <sub>1</sub> X <sub>3</sub> X <sub>4</sub>	10	255.3	25.53	0.13	0.99	1.61	X <sub>1</sub> X <sub>3</sub> X <sub>4</sub>	10	12.89	1.28	0.48	0.9	3.66
X <sub>1</sub> X <sub>2</sub> X <sub>3</sub> X <sub>4</sub>	5	132.93	26.58	0.13	0.98	0.83	X <sub>1</sub> X <sub>2</sub> X <sub>3</sub> X <sub>4</sub>	5	0.66	0.13	0.05	0.99	0.18
Total	58	15845.9	-	-	-	100	Total	58	351.41	-	-	-	100

Factors: coagulant dosage (mg/l; X<sub>1</sub>), initial pH (X<sub>2</sub>), type of chemical coagulant (X<sub>3</sub>) and type of natural coagulant (X<sub>4</sub>)

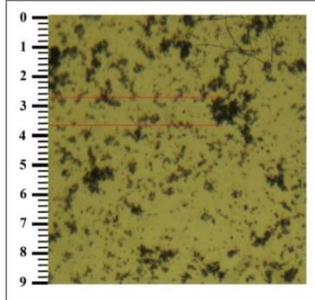
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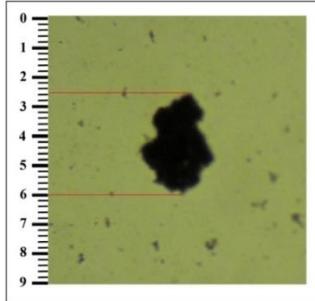
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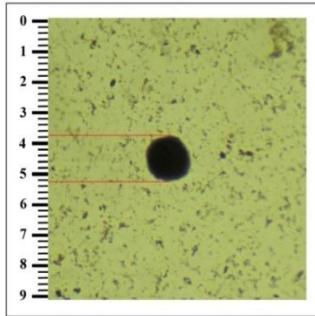
PAC -Dolomit



Alum -Dolomit

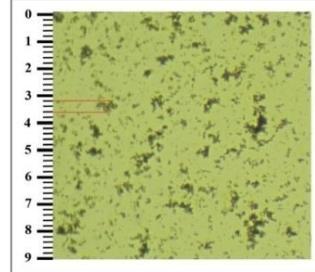


FeCl<sub>3</sub> -Dolomit

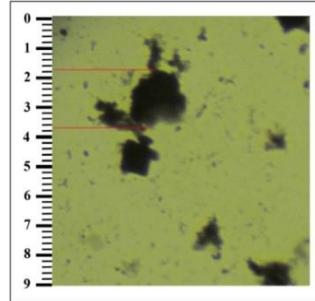


Microscope Images of the flocs (mm) produced by 3 mg/l PAC + dolomite, 3 mg/l PAC + wheat starch, 3 mg/l alum + dolomite, 3 mg/l alum + wheat starch, 3 mg/l FeCl<sub>3</sub> + dolomite and 3 mg/l FeCl<sub>3</sub> + wheat starch at 30 min sedimentation time

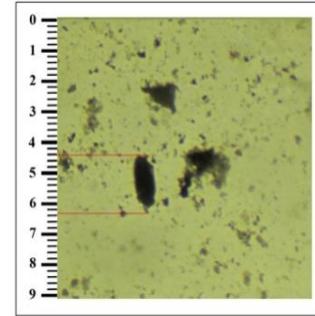
Raw Wastewater



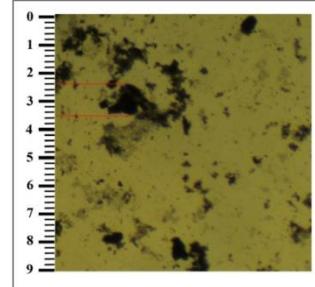
PAC -Starch



Alum - Starch



FeCl<sub>3</sub> - Starch



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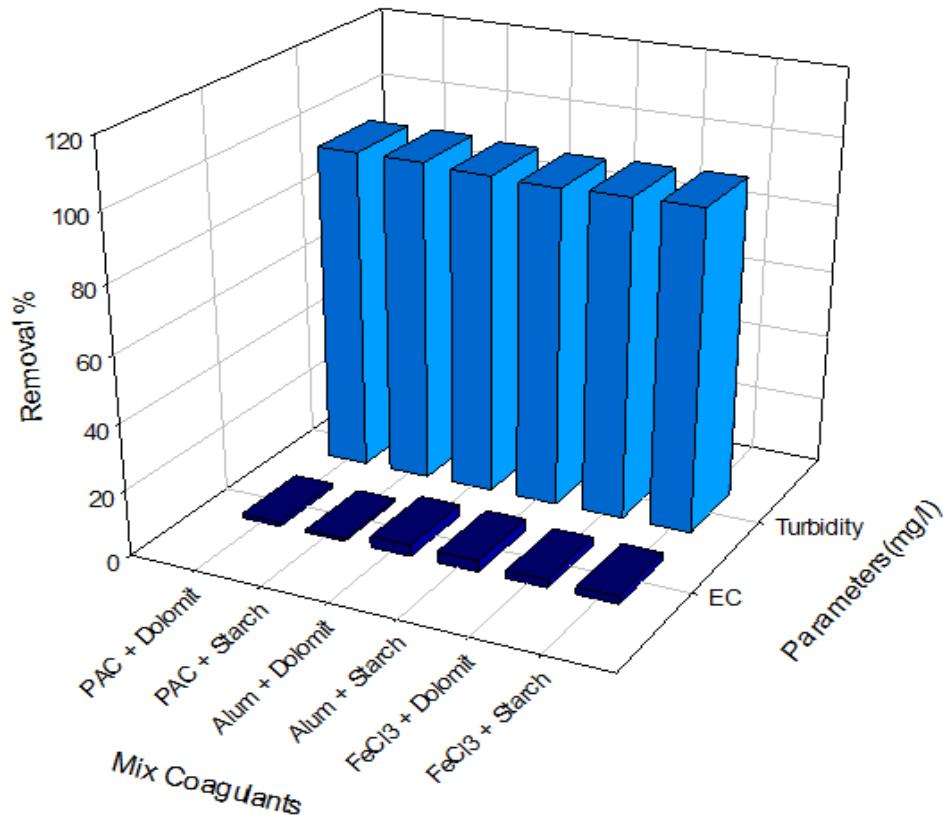
**Discussion**

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- The results of this study showed the potential use of dolomite powder and wheat starch as primary or partial natural coagulants to decrease the negative contributions of conventional chemical coagulants.
- This study indicated the potential of using alum and  $\text{FeCl}_3$  as a primary chemical coagulant or coagulant aid to reduce the purchase costs of chemical coagulants that should be used in the wastewater treatment process.
- Further study on the cost and benefits and life cycle analysis should be conducted to better utilize wheat starch and dolomite as inexpensive and viable natural sources and to better utilize alum and  $\text{FeCl}_3$  as a cheap and efficient chemical coagulant source for wastewater treatment at blast furnace 3 of Isfahan Steel Company.



**Thank You!**